

# Phenol, 2-(trifluoromethyl)-

<b>Other names:</b>	o-Cresol, «alpha», «alpha», «alpha»-trifluoro-o-(Trifluoromethoxy)phenol o-Hydroxybenzotrifluoride 2-(Trifluoromethyl)phenol «alpha», «alpha», «alpha»-Trifluoro-o-cresol 2-Hydroxybenzotrifluoride o-(Trifluoromethyl)phenol
<b>Inchi:</b>	InChI=1S/C7H5F3O/c8-7(9,10)5-3-1-2-4-6(5)11/h1-4,11H
<b>InchiKey:</b>	ZOQOPXVJANRGJZ-UHFFFAOYSA-N
<b>Formula:</b>	C7H5F3O
<b>SMILES:</b>	Oc1ccccc1C(F)(F)F
<b>Mol. weight [g/mol]:</b>	162.11
<b>CAS:</b>	444-30-4

## Physical Properties

Property code	Value	Unit	Source
gf	-615.74	kJ/mol	Joback Method
hf	-725.67	kJ/mol	Joback Method
hfus	15.54	kJ/mol	Joback Method
hvap	42.72	kJ/mol	Joback Method
log10ws	-2.12		Crippen Method
logp	2.411		Crippen Method
mvol	96.910	ml/mol	McGowan Method
pc	4294.29	kPa	Joback Method
tb	420.70	K	NIST Webbook
tb	420.50 ± 0.50	K	NIST Webbook
tc	669.45	K	Joback Method
tf	310.98	K	Joback Method
vc	0.329	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	212.51	J/mol×K	461.44	Joback Method

cpg	222.79	J/mol×K	496.11	Joback Method
cpg	232.14	J/mol×K	530.78	Joback Method
cpg	240.62	J/mol×K	565.44	Joback Method
cpg	248.33	J/mol×K	600.11	Joback Method
cpg	255.34	J/mol×K	634.78	Joback Method
cpg	261.74	J/mol×K	669.45	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C444304&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C444304&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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