

# Dimethylmalonic acid, heptyl 3-methylphenyl ester

Inchi:	InChI=1S/C19H28O4/c1-5-6-7-8-9-13-22-17(20)19(3,4)18(21)23-16-12-10-11-15(2)14-16
InchiKey:	DGBOTRJUYMCQLN-UHFFFAOYSA-N
Formula:	C19H28O4
SMILES:	CCCCCCCOC(=O)C(C)(C)C(=O)Oc1cccc(C)c1
Mol. weight [g/mol]:	320.42

## Physical Properties

Property code	Value	Unit	Source
gf	-253.12	kJ/mol	Joback Method
hf	-708.78	kJ/mol	Joback Method
hfus	36.78	kJ/mol	Joback Method
hvap	77.84	kJ/mol	Joback Method
log10ws	-5.07		Crippen Method
logp	4.440		Crippen Method
mcvol	269.690	ml/mol	McGowan Method
pc	1462.37	kPa	Joback Method
rinpol	2108.00		NIST Webbook
tb	815.13	K	Joback Method
tc	1020.72	K	Joback Method
tf	489.57	K	Joback Method
vc	1.028	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	820.35	J/molxK	815.13	Joback Method
cpg	836.47	J/molxK	849.40	Joback Method
cpg	851.47	J/molxK	883.66	Joback Method
cpg	865.37	J/molxK	917.93	Joback Method
cpg	878.23	J/molxK	952.19	Joback Method
cpg	890.08	J/molxK	986.46	Joback Method
cpg	900.96	J/molxK	1020.72	Joback Method
dvisc	0.0006091	Paxs	489.57	Joback Method
dvisc	0.0003266	Paxs	543.83	Joback Method

dvisc	0.0001960	Paxs	598.09	Joback Method
dvisc	0.0001281	Paxs	652.35	Joback Method
dvisc	0.0000894	Paxs	706.61	Joback Method
dvisc	0.0000656	Paxs	760.87	Joback Method
dvisc	0.0000502	Paxs	815.13	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U363615&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U363615&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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