

# 1,3-Benzenediamine, 4-chloro-

<b>Other names:</b>	m-Phenylenediamine, 4-chloro- C.I. 76027 1-Chloro-2,4-Diaminobenzene 4-Chloro-m-phenylenediamine 4-Chloro-1,3-benzenediamine 4-Chloro-1,3-phenylenediamine NCI-C03305 4-Chloro-meta-phenylenediamine 4-Chlorophene-1,3-diamine 4-Chlorophenylene-1,3-diamine p-Chlor-m-fenylendiamin p-Chloro-m-phenylenediamine 4-Cl-m-Pd 3-Amino-4-chloroaniline NSC 6074 4-chlorobenzene-1,3-diamine 2,4-Diaminochlorobenzene
<b>Inchi:</b>	InChI=1S/C6H7CIN2/c7-5-2-1-4(8)3-6(5)9/h1-3H,8-9H2
<b>InchiKey:</b>	ZWUBBMDHSZDNTA-UHFFFAOYSA-N
<b>Formula:</b>	C6H7CIN2
<b>SMILES:</b>	<chem>Nc1ccc(Cl)c(N)c1</chem>
<b>Mol. weight [g/mol]:</b>	142.59
<b>CAS:</b>	5131-60-2

## Physical Properties

Property code	Value	Unit	Source
gf	213.76	kJ/mol	Joback Method
hf	98.26	kJ/mol	Joback Method
hfus	19.15	kJ/mol	Joback Method
hvap	58.22	kJ/mol	Joback Method
log10ws	-1.42		Crippen Method
logp	1.504		Crippen Method
mcvol	103.840	ml/mol	McGowan Method
pc	4987.39	kPa	Joback Method
tb	555.81	K	Joback Method
tc	805.36	K	Joback Method
tf	405.28	K	Joback Method

vc

0.370

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	216.19	J/mol×K	555.81	Joback Method
cpg	225.31	J/mol×K	597.40	Joback Method
cpg	233.79	J/mol×K	638.99	Joback Method
cpg	241.64	J/mol×K	680.58	Joback Method
cpg	248.91	J/mol×K	722.17	Joback Method
cpg	255.62	J/mol×K	763.77	Joback Method
cpg	261.79	J/mol×K	805.36	Joback Method

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C5131602&Units=SI>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point

**vc:** Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/69-280-5/1-3-Benzenediamine-4-chloro.pdf>

Generated by Cheméo on 2024-04-25 17:07:57.68624367 +0000 UTC m=+16354126.606820992.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.