

# p-N,N-Diethylaminostyrene

<b>Inchi:</b>	InChI=1S/C12H17N/c1-4-11-7-9-12(10-8-11)13(5-2)6-3/h4,7-10H,1,5-6H2,2-3H3
<b>InchiKey:</b>	CBZMQWPBAUBAPO-UHFFFAOYSA-N
<b>Formula:</b>	C12H17N
<b>SMILES:</b>	C=Cc1ccc(N(CC)CC)cc1
<b>Mol. weight [g/mol]:</b>	175.27

## Physical Properties

Property code	Value	Unit	Source
gf	351.56	kJ/mol	Joback Method
hf	127.01	kJ/mol	Joback Method
hfus	22.23	kJ/mol	Joback Method
hvap	46.62	kJ/mol	Joback Method
log10ws	-3.04		Crippen Method
logp	3.176		Crippen Method
mcvol	161.860	ml/mol	McGowan Method
pc	2465.36	kPa	Joback Method
ripol	1523.80		NIST Webbook
ripol	1524.30		NIST Webbook
ripol	1523.80		NIST Webbook
ripol	2005.20		NIST Webbook
ripol	2005.80		NIST Webbook
ripol	2005.20		NIST Webbook
tb	514.74	K	Joback Method
tc	716.70	K	Joback Method
tf	294.65	K	Joback Method
vc	0.599	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	362.15	J/molxK	514.74	Joback Method
cpg	378.76	J/molxK	548.40	Joback Method
cpg	394.40	J/molxK	582.06	Joback Method
cpg	409.14	J/molxK	615.72	Joback Method

cpg	423.01	J/mol×K	649.38	Joback Method
cpg	436.05	J/mol×K	683.04	Joback Method
cpg	448.31	J/mol×K	716.70	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R246509&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R246509&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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