

# Benzyl methyl sulfide

<b>Other names:</b>	Benzene, [(methylthio)methyl]- Sulfide, benzyl methyl Methyl benzyl sulfide 1-Phenyl-2-thiapropene Methylthiomethylbenzene «alpha»-(Methylthio)toluene Benzyl methyl sulphide NSC 75125
<b>Inchi:</b>	InChI=1S/C8H10S/c1-9-7-8-5-3-2-4-6-8/h2-6H,7H2,1H3
<b>InchiKey:</b>	OFQPKKGMNWASPN-UHFFFAOYSA-N
<b>Formula:</b>	C8H10S
<b>SMILES:</b>	CSCc1ccccc1
<b>Mol. weight [g/mol]:</b>	138.23
<b>CAS:</b>	766-92-7

## Physical Properties

Property code	Value	Unit	Source
chl	-5205.10 ± 2.00	kJ/mol	NIST Webbook
gf	162.01	kJ/mol	Joback Method
hf	79.00 ± 3.00	kJ/mol	NIST Webbook
hfl	26.00 ± 2.00	kJ/mol	NIST Webbook
hfus	14.65	kJ/mol	Joback Method
hvap	54.00 ± 2.00	kJ/mol	NIST Webbook
hvap	53.00	kJ/mol	NIST Webbook
hvap	55.20 ± 2.10	kJ/mol	NIST Webbook
ie	8.42	eV	NIST Webbook
ie	8.20	eV	NIST Webbook
ie	8.64	eV	NIST Webbook
ie	8.41	eV	NIST Webbook
log10ws	-2.65		Crippen Method
logp	2.550		Crippen Method
mcvol	116.170	ml/mol	McGowan Method
pc	3727.11	kPa	Joback Method
rinpol	1176.00		NIST Webbook
rinpol	1200.00		NIST Webbook
rinpol	1131.00		NIST Webbook
rinpol	1147.00		NIST Webbook

ripol	1184.00		NIST Webbook
ripol	1185.00		NIST Webbook
ripol	1176.00		NIST Webbook
ripol	1176.00		NIST Webbook
ripol	1167.00		NIST Webbook
ripol	1183.00		NIST Webbook
ripol	1680.00		NIST Webbook
ripol	1664.00		NIST Webbook
ripol	1665.00		NIST Webbook
ripol	1665.00		NIST Webbook
tb	469.70	K	NIST Webbook
tc	712.59	K	Joback Method
tf	240.74	K	Joback Method
vc	0.429	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	222.99	J/mol×K	477.90	Joback Method
cpg	236.28	J/mol×K	517.01	Joback Method
cpg	248.73	J/mol×K	556.13	Joback Method
cpg	260.36	J/mol×K	595.24	Joback Method
cpg	271.21	J/mol×K	634.36	Joback Method
cpg	281.31	J/mol×K	673.47	Joback Method
cpg	290.68	J/mol×K	712.59	Joback Method
hvapt	51.80	kJ/mol	352.00	NIST Webbook
hvapt	50.80	kJ/mol	352.00	NIST Webbook

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C766927&Units=SI>

# Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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