

# Benzene, 1-butyl-4-(2-methylpropyl)

Inchi:	InChI=1S/C14H22/c1-4-5-6-13-7-9-14(10-8-13)11-12(2)3/h7-10,12H,4-6,11H2,1-3H3
InchiKey:	YGZJMJYMAYHEMM-UHFFFAOYSA-N
Formula:	C14H22
SMILES:	CCCCc1ccc(CC(C)C)cc1
Mol. weight [g/mol]:	190.32

## Physical Properties

Property code	Value	Unit	Source
gf	167.34	kJ/mol	Joback Method
hf	-112.51	kJ/mol	Joback Method
hfus	22.15	kJ/mol	Joback Method
hvap	49.31	kJ/mol	Joback Method
log10ws	-4.51		Crippen Method
logp	4.228		Crippen Method
mcvol	184.360	ml/mol	McGowan Method
pc	1996.55	kPa	Joback Method
ripol	1360.00		NIST Webbook
ripol	1360.00		NIST Webbook
ripol	1561.00		NIST Webbook
ripol	1569.20		NIST Webbook
ripol	1561.00		NIST Webbook
ripol	1561.00		NIST Webbook
tb	550.94	K	Joback Method
tc	750.04	K	Joback Method
tf	271.48	K	Joback Method
vc	0.706	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	442.71	J/molxK	550.94	Joback Method
cpg	461.01	J/molxK	584.12	Joback Method
cpg	478.35	J/molxK	617.31	Joback Method
cpg	494.77	J/molxK	650.49	Joback Method

cpg	510.30	J/molxK	683.67	Joback Method
cpg	524.99	J/molxK	716.85	Joback Method
cpg	538.86	J/molxK	750.04	Joback Method
dvisc	0.0035872	Paxs	271.48	Joback Method
dvisc	0.0014741	Paxs	318.06	Joback Method
dvisc	0.0007603	Paxs	364.63	Joback Method
dvisc	0.0004556	Paxs	411.21	Joback Method
dvisc	0.0003030	Paxs	457.79	Joback Method
dvisc	0.0002173	Paxs	504.36	Joback Method
dvisc	0.0001648	Paxs	550.94	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R12579&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R12579&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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