

# Tetramethylene chloriodide

<b>Other names:</b>	1-Chloro-4-iodobutane Butane, 1-chloro-4-iodo-
<b>Inchi:</b>	InChI=1S/C4H8ClI/c5-3-1-2-4-6/h1-4H2
<b>InchiKey:</b>	JXOSPTBRSOYXGC-UHFFFAOYSA-N
<b>Formula:</b>	C4H8ClI
<b>SMILES:</b>	CICCCCI
<b>Mol. weight [g/mol]:</b>	218.46
<b>CAS:</b>	10297-05-9

## Physical Properties

Property code	Value	Unit	Source
gf	28.99	kJ/mol	Joback Method
hf	-64.76	kJ/mol	Joback Method
hfus	14.72	kJ/mol	Joback Method
hvap	38.26	kJ/mol	Joback Method
log10ws	-2.60		Crippen Method
logp	2.440		Crippen Method
mvol	105.280	ml/mol	McGowan Method
pc	3624.61	kPa	Joback Method
tb	421.49	K	Joback Method
tc	633.80	K	Joback Method
tf	222.82	K	Joback Method
vc	0.397	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	158.78	J/mol×K	421.49	Joback Method
cpg	193.34	J/mol×K	598.42	Joback Method
cpg	187.23	J/mol×K	563.03	Joback Method
cpg	180.74	J/mol×K	527.65	Joback Method
cpg	173.85	J/mol×K	492.26	Joback Method
cpg	166.54	J/mol×K	456.88	Joback Method
cpg	199.09	J/mol×K	633.80	Joback Method

dvisc	0.0004224	Paxs	421.49	Joback Method
dvisc	0.0005400	Paxs	388.38	Joback Method
dvisc	0.0007225	Paxs	355.27	Joback Method
dvisc	0.0010265	Paxs	322.15	Joback Method
dvisc	0.0015806	Paxs	289.04	Joback Method
dvisc	0.0027214	Paxs	255.93	Joback Method
dvisc	0.0055064	Paxs	222.82	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10297059&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10297059&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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