

# 3-Cyclopentylpropionic acid, pentyl ester

<b>Inchi:</b>	InChI=1S/C13H24O2/c1-2-3-6-11-15-13(14)10-9-12-7-4-5-8-12/h12H,2-11H2,1H3
<b>InchiKey:</b>	JZKZDGPVKBEGPD-UHFFFAOYSA-N
<b>Formula:</b>	C13H24O2
<b>SMILES:</b>	CCCCCOC(=O)CCC1CCCC1
<b>Mol. weight [g/mol]:</b>	212.33

## Physical Properties

Property code	Value	Unit	Source
gf	-138.79	kJ/mol	Joback Method
hf	-495.97	kJ/mol	Joback Method
hfus	26.15	kJ/mol	Joback Method
hvap	53.94	kJ/mol	Joback Method
log10ws	-3.78		Crippen Method
logp	3.690		Crippen Method
mvol	190.610	ml/mol	McGowan Method
pc	1992.98	kPa	Joback Method
rinpol	1566.00		NIST Webbook
tb	588.41	K	Joback Method
tc	779.04	K	Joback Method
tf	319.33	K	Joback Method
vc	0.729	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	504.67	J/molxK	588.41	Joback Method
cpg	523.31	J/molxK	620.18	Joback Method
cpg	541.03	J/molxK	651.95	Joback Method
cpg	557.84	J/molxK	683.72	Joback Method
cpg	573.76	J/molxK	715.50	Joback Method
cpg	588.83	J/molxK	747.27	Joback Method
cpg	603.06	J/molxK	779.04	Joback Method
dvisc	0.0031696	Paxs	319.33	Joback Method
dvisc	0.0015749	Paxs	364.18	Joback Method

dvisc	0.0009122	Paxs	409.02	Joback Method
dvisc	0.0005886	Paxs	453.87	Joback Method
dvisc	0.0004109	Paxs	498.72	Joback Method
dvisc	0.0003044	Paxs	543.56	Joback Method
dvisc	0.0002360	Paxs	588.41	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U292332&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U292332&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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