

# Oxirane, trimethyl-

<b>Other names:</b>	Butane, 2,3-epoxy-2-methyl- «beta»-Isoamylene oxide Trimethylethylene oxide Trimethyloxacyclopropane Trimethyloxirane 2-Methyl-2-butene oxide 2-Methyl-2,3-epoxybutane 2,3-Epoxy-2-methylbutane oxirane, 2,2,3-trimethyl- 2,2,3-Trimethyloxirane «alpha»-Trimethylethylene oxide
<b>Inchi:</b>	InChI=1S/C5H10O/c1-4-5(2,3)6-4/h4H,1-3H3
<b>InchiKey:</b>	QPBYBLZYMNWGMU-UHFFFAOYSA-N
<b>Formula:</b>	C5H10O
<b>SMILES:</b>	CC1OC1(C)C
<b>Mol. weight [g/mol]:</b>	86.13
<b>CAS:</b>	5076-19-7

## Physical Properties

Property code	Value	Unit	Source
chl	-3133.00	kJ/mol	NIST Webbook
gf	-47.35	kJ/mol	Joback Method
hf	-210.83	kJ/mol	Joback Method
hfus	9.59	kJ/mol	Joback Method
hvap	29.69	kJ/mol	Joback Method
log10ws	-1.12		Crippen Method
logp	1.184		Crippen Method
mcvol	76.320	ml/mol	McGowan Method
pc	4082.92	kPa	Joback Method
rinpol	668.00		NIST Webbook
rinpol	668.20		NIST Webbook
rinpol	668.00		NIST Webbook
rinpol	649.00		NIST Webbook
rinpol	668.20		NIST Webbook
rinpol	668.10		NIST Webbook
rinpol	668.00		NIST Webbook
rinpol	668.00		NIST Webbook

rinpol	668.00		NIST Webbook
tb	348.65 ± 2.00	K	NIST Webbook
tc	531.54	K	Joback Method
tf	210.28	K	Joback Method
vc	0.290	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	134.28	J/mol×K	343.06	Joback Method
cpg	146.27	J/mol×K	374.47	Joback Method
cpg	157.32	J/mol×K	405.89	Joback Method
cpg	167.52	J/mol×K	437.30	Joback Method
cpg	176.93	J/mol×K	468.71	Joback Method
cpg	185.65	J/mol×K	500.12	Joback Method
cpg	193.73	J/mol×K	531.54	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C5076197&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C5076197&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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