

# (E,E,Z)-1,5,9-Cyclododecatriene, 1-methyl

<b>Inchi:</b>	InChI=1S/C13H20/c1-13-11-9-7-5-3-2-4-6-8-10-12-13/h3,5-6,8,11H,2,4,7,9-10,12H2,1H3
<b>InchiKey:</b>	VJHWJWRAAFGNQY-QQRKSBPLSA-N
<b>Formula:</b>	C13H20
<b>SMILES:</b>	CC1=CCCC=CCCC=CCC1
<b>Mol. weight [g/mol]:</b>	176.30

## Physical Properties

Property code	Value	Unit	Source
gf	98.39	kJ/mol	Joback Method
hf	-112.08	kJ/mol	Joback Method
hfus	10.87	kJ/mol	Joback Method
hvap	47.84	kJ/mol	Joback Method
log10ws	-4.72		Crippen Method
logp	4.399		Crippen Method
mcvol	170.270	ml/mol	McGowan Method
pc	2537.93	kPa	Joback Method
rinpol	1364.00		NIST Webbook
tb	549.14	K	Joback Method
tc	793.41	K	Joback Method
tf	241.57	K	Joback Method
vc	0.608	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	398.51	J/molxK	549.14	Joback Method
cpg	422.75	J/molxK	589.85	Joback Method
cpg	445.48	J/molxK	630.56	Joback Method
cpg	466.70	J/molxK	671.27	Joback Method
cpg	486.40	J/molxK	711.99	Joback Method
cpg	504.59	J/molxK	752.70	Joback Method
cpg	521.25	J/molxK	793.41	Joback Method
dvisc	0.0352747	Paxs	241.57	Joback Method
dvisc	0.0044782	Paxs	292.83	Joback Method

dvisc	0.0010515	Paxs	344.09	Joback Method
dvisc	0.0003595	Paxs	395.36	Joback Method
dvisc	0.0001573	Paxs	446.62	Joback Method
dvisc	0.0000816	Paxs	497.88	Joback Method
dvisc	0.0000478	Paxs	549.14	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R2781&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R2781&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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