

# 1-Hexene, 1-iodo-, (Z)-

<b>Other names:</b>	cis-1-Iodo-1-Hexene cis-1-Iodohexene
<b>Inchi:</b>	InChI=1S/C6H11I/c1-2-3-4-5-6-7/h5-6H,2-4H2,1H3/b6-5-
<b>InchiKey:</b>	TUAPXMNNQGXDDV-WAYWQWQTSA-N
<b>Formula:</b>	C6H11I
<b>SMILES:</b>	CCCCC=CI
<b>Mol. weight [g/mol]:</b>	210.06
<b>CAS:</b>	16538-47-9

## Physical Properties

Property code	Value	Unit	Source
gf	137.98	kJ/mol	Joback Method
hf	26.92	kJ/mol	Joback Method
hfus	15.90	kJ/mol	Joback Method
hvap	38.28	kJ/mol	Joback Method
log10ws	-3.63		Crippen Method
logp	3.125		Crippen Method
mvol	116.920	ml/mol	McGowan Method
pc	3213.68	kPa	Joback Method
tb	433.98	K	Joback Method
tc	644.76	K	Joback Method
tf	210.36	K	Joback Method
vc	0.440	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.53	J/molxK	433.98	Joback Method
cpg	204.09	J/molxK	469.11	Joback Method
cpg	214.00	J/molxK	504.24	Joback Method
cpg	223.32	J/molxK	539.37	Joback Method
cpg	232.07	J/molxK	574.50	Joback Method
cpg	240.29	J/molxK	609.63	Joback Method
cpg	248.02	J/molxK	644.76	Joback Method

dvisc	0.0058545	Paxs	210.36	Joback Method
dvisc	0.0024684	Paxs	247.63	Joback Method
dvisc	0.0013046	Paxs	284.90	Joback Method
dvisc	0.0007992	Paxs	322.17	Joback Method
dvisc	0.0005419	Paxs	359.44	Joback Method
dvisc	0.0003953	Paxs	396.71	Joback Method
dvisc	0.0003044	Paxs	433.98	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.81197e+01
Coeff. B	-4.86276e+03
Coeff. C	-6.46240e+01
Temperature range (K), min.	337.32
Temperature range (K), max.	444.28

## Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C16538479&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C16538479&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>p<sub>vap</sub>:</b>	Vapor pressure
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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