

Acetic acid, sodium salt

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|-----------------------------|--|
| Other names: | anhydrous sodium acetate sodium acetate sodium ethanoate |
| Inchi: | InChI=1S/C2H4O2.Na/c1-2(3)4;/h1H3,(H,3,4);/q;+1/p-1 |
| InchiKey: | VMHLLURERBWHNL-UHFFFAOYSA-M |
| Formula: | C2H3NaO2 |
| SMILES: | CC(=O)O[Na] |
| Mol. weight [g/mol]: | 82.03 |
| CAS: | 127-09-3 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------------|---------|---|
| hfus | 17.40 | kJ/mol | Enthalpies of formation and lattice enthalpies of alkaline metal acetates |
| ss | 138.10 | J/mol×K | NIST Webbook |
| ss | 123.09 | J/mol×K | NIST Webbook |
| tf | 601.30 ± 0.20 | K | NIST Webbook |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|--------------|
| cps | 100.83 | J/mol×K | 298.15 | NIST Webbook |
| cps | 111.70 | J/mol×K | 340.00 | NIST Webbook |
| cps | 88.07 | J/mol×K | 291.18 | NIST Webbook |

Sources

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 - <https://www.doi.org/10.1021/je300701m>
 - <https://www.doi.org/10.1021/je500419b>
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Legend

- cps: Solid phase heat capacity
- hfus: Enthalpy of fusion at standard conditions
- ss: Solid phase molar entropy at standard conditions
- tf: Normal melting (fusion) point

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