

Acetic acid, sodium salt

Other names:	anhydrous sodium acetate
	sodium acetate
	sodium ethanoate
Inchi:	InChI=1S/C2H4O2.Na/c1-2(3)4;/h1H3,(H,3,4);/q;+1/p-1
InchiKey:	VMHLLURERBWHNL-UHFFFAOYSA-M
Formula:	C2H3NaO2
SMILES:	CC(=O)O[Na]
Mol. weight [g/mol]:	82.03
CAS:	127-09-3

Physical Properties

Property code	Value	Unit	Source
hfus	17.40	kJ/mol	Enthalpies of formation and lattice enthalpies of alkaline metal acetates
ss	138.10	J/molxK	NIST Webbook
ss	123.09	J/molxK	NIST Webbook
tf	601.30 ± 0.20	K	NIST Webbook

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cps	100.83	J/molxK	298.15	NIST Webbook
cps	111.70	J/molxK	340.00	NIST Webbook
cps	88.07	J/molxK	291.18	NIST Webbook

Sources

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 Activity of Water and Organic Solvents in Aqueous Basic Solutions
 Coefficients of water and basic solvents in vapor-liquid equilibria for water + acetic acid + sodium acetate
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 Equilibria in the Ternary System NaCl NaAc H₂O at 303.15 K, 323.15 K, and 343.15 K
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 Aqueous Two-Phase Systems and Solid Phase Formation
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Legend

cps: Solid phase heat capacity
 hfus: Enthalpy of fusion at standard conditions
 ss: Solid phase molar entropy at standard conditions
 tf: Normal melting (fusion) point

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