

# 1,3,3-Trimethylcyclopropene

<b>Other names:</b>	Cyclopropene-1,3,3-trimethyl
<b>Inchi:</b>	InChI=1S/C6H10/c1-5-4-6(5,2)3/h4H,1-3H3
<b>InchiKey:</b>	MRLAULBRWNYTDV-UHFFFAOYSA-N
<b>Formula:</b>	C6H10
<b>SMILES:</b>	CC1=CC1(C)C
<b>Mol. weight [g/mol]:</b>	82.14
<b>CAS:</b>	3664-56-0

## Physical Properties

Property code	Value	Unit	Source
affp	895.40	kJ/mol	NIST Webbook
basg	865.90	kJ/mol	NIST Webbook
gf	75.23	kJ/mol	Joback Method
hf	-32.82	kJ/mol	Joback Method
hfus	3.97	kJ/mol	Joback Method
hvap	26.80 ± 1.70	kJ/mol	NIST Webbook
hvap	27.90 ± 1.40	kJ/mol	NIST Webbook
ie	8.58 ± 0.05	eV	NIST Webbook
ie	9.00	eV	NIST Webbook
ie	8.70	eV	NIST Webbook
log10ws	-1.84		Crippen Method
logp	1.973		Crippen Method
mcvol	80.240	ml/mol	McGowan Method
pc	3916.03	kPa	Joback Method
tb	347.80	K	Joback Method
tc	538.54	K	Joback Method
tf	212.50	K	Joback Method
vc	0.312	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	133.06	J/mol×K	347.80	Joback Method
cpg	144.66	J/mol×K	379.59	Joback Method

cpg	155.30	J/mol×K	411.38	Joback Method
cpg	165.05	J/mol×K	443.17	Joback Method
cpg	174.00	J/mol×K	474.96	Joback Method
cpg	182.24	J/mol×K	506.75	Joback Method
cpg	189.85	J/mol×K	538.54	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3664560&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3664560&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/69-906-0/1-3-3-Trimethylcyclopropene.pdf>

Generated by Cheméo on 2024-04-23 15:28:43.839091033 +0000 UTC m=+16175372.759668348.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.