

# 1,1-Dimethoxyethane

Inchi:	InChI=1S/C4H10O2/c1-4(5-2)6-3/h4H,1-3H3
InchiKey:	SPEUIVXLLWOEMJ-UHFFFAOYSA-N
Formula:	C4H10O2
SMILES:	COC(C)OC
Mol. weight [g/mol]:	90.12
CAS:	25154-53-4

## Physical Properties

Property code	Value	Unit	Source
chg	-2613.50 ± 0.79	kJ/mol	NIST Webbook
gf	-229.64	kJ/mol	Joback Method
hf	-389.70 ± 0.84	kJ/mol	NIST Webbook
hfl	-419.99 ± 0.84	kJ/mol	NIST Webbook
hfus	4.97	kJ/mol	Joback Method
hvap	28.93	kJ/mol	Joback Method
log10ws	-0.28		Crippen Method
logp	0.625		Crippen Method
mcvol	78.960	ml/mol	McGowan Method
pc	3773.04	kPa	Joback Method
tb	335.32	K	Joback Method
tc	504.58	K	Joback Method
tf	164.30	K	Joback Method
vc	0.289	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	136.91	J/molxK	335.32	Joback Method
cpg	171.73	J/molxK	476.37	Joback Method
cpg	164.99	J/molxK	448.16	Joback Method
cpg	158.12	J/molxK	419.95	Joback Method
cpg	151.14	J/molxK	391.74	Joback Method
cpg	144.07	J/molxK	363.53	Joback Method
cpg	178.35	J/molxK	504.58	Joback Method

dvisc	0.0001930	Paxs	335.32	Joback Method
dvisc	0.0002524	Paxs	306.82	Joback Method
dvisc	0.0003489	Paxs	278.31	Joback Method
dvisc	0.0005190	Paxs	249.81	Joback Method
dvisc	0.0008554	Paxs	221.31	Joback Method
dvisc	0.0016342	Paxs	192.80	Joback Method
dvisc	0.0039085	Paxs	164.30	Joback Method

## Sources

<b>KDB Vapor Pressure Data:</b>	<a href="https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1049">https://www.cheric.org/research/kdb/hcprop/showprop.php?cmpid=1049</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C25154534&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C25154534&amp;Units=SI</a>

## Legend

<b>chg:</b>	Standard gas enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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