

# Benzene, 1-chloro-3,5-bis(1,1-dimethylethyl)-

<b>Other names:</b>	3,5-Di-tert-butyl chlorobenzene
<b>Inchi:</b>	InChI=1S/C14H21Cl/c1-13(2,3)10-7-11(14(4,5)6)9-12(15)8-10/h7-9H,1-6H3
<b>InchiKey:</b>	UMPNQCBQSXTKID-UHFFFAOYSA-N
<b>Formula:</b>	C14H21Cl
<b>SMILES:</b>	CC(C)(C)c1cc(Cl)cc(C(C)(C)C)c1
<b>Mol. weight [g/mol]:</b>	224.77
<b>CAS:</b>	80438-67-1

## Physical Properties

Property code	Value	Unit	Source
gf	153.90	kJ/mol	Joback Method
hf	-151.94	kJ/mol	Joback Method
hfl	-229.00 ± 2.00	kJ/mol	NIST Webbook
hfus	14.65	kJ/mol	Joback Method
hvap	52.15	kJ/mol	Joback Method
log10ws	-4.78		Crippen Method
logp	4.935		Crippen Method
mcvol	196.600	ml/mol	McGowan Method
pc	1957.87	kPa	Joback Method
tb	587.33	K	Joback Method
tc	812.50	K	Joback Method
tf	333.76	K	Joback Method
vc	0.739	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	480.72	J/molxK	587.33	Joback Method
cpg	561.64	J/molxK	774.98	Joback Method
cpg	547.73	J/molxK	737.45	Joback Method
cpg	532.77	J/molxK	699.92	Joback Method
cpg	516.67	J/molxK	662.39	Joback Method
cpg	499.35	J/molxK	624.86	Joback Method
cpg	574.60	J/molxK	812.50	Joback Method

dvisc	0.0001357	Paxs	587.33	Joback Method
dvisc	0.0001818	Paxs	545.07	Joback Method
dvisc	0.0002557	Paxs	502.81	Joback Method
dvisc	0.0003828	Paxs	460.54	Joback Method
dvisc	0.0006219	Paxs	418.28	Joback Method
dvisc	0.0011269	Paxs	376.02	Joback Method
dvisc	0.0023734	Paxs	333.76	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.11942e+01
Coeff. B	-3.33206e+03
Coeff. C	-8.84900e+01
Temperature range (K), min.	394.00
Temperature range (K), max.	654.91

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C80438671&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C80438671&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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