

# «alpha»-Ionone

<b>Other names:</b>	3-Buten-2-one, 4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-, (E)-«alpha»-Cyclocitrylideneacetone (E)-«alpha»-Ionone trans-«alpha»-Ionone 3-Buten-2-one, 4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-, trans 4-(2,6,6-Trimethyl-2-cyclohexen-1-yl)-3-buten-2-one, (E)- (3E)-4-(2,6,6-Trimethyl-2-cyclohexen-1-yl)-3-buten-2-one «alpha»-(E)-Ionone trans-4-(2,6,6-Trimethyl-2-cyclohexen-1-yl)-3-buten-2-one 3-Buten-2-one, 4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-, (3E)- 4-(2,6,6-trimethyl-2-cyclohexen-1-yl)-3-buten-2-one («alpha»-ionone) 4-(2,6,6-Trimethyl-2-cyclohexen-1-yl)-3-buten-2-one 4-(2,6,6-trimethylcyclohex-2-ene-1-yl)-but-3-ene-2-one
<b>Inchi:</b>	InChI=1S/C13H20O/c1-10-6-5-9-13(3,4)12(10)8-7-11(2)14/h6-8,12H,5,9H2,1-4H3/b8-7+
<b>InchiKey:</b>	UZFLPKAIBPNNCA-BQYQJAHWSA-N
<b>Formula:</b>	C13H20O
<b>SMILES:</b>	CC(=O)C=CC1C(C)=CCCC1(C)C
<b>Mol. weight [g/mol]:</b>	192.30
<b>CAS:</b>	127-41-3

## Physical Properties

Property code	Value	Unit	Source
gf	41.46	kJ/mol	Joback Method
hf	-211.48	kJ/mol	Joback Method
hfus	18.67	kJ/mol	Joback Method
hvap	51.16	kJ/mol	Joback Method
log10ws	-3.66		Crippen Method
logp	3.514		Crippen Method
mcvol	176.140	ml/mol	McGowan Method
pc	2239.76	kPa	Joback Method
rinpol	1426.00		NIST Webbook
rinpol	1429.00		NIST Webbook
rinpol	1410.00		NIST Webbook
rinpol	1412.00		NIST Webbook
rinpol	1416.00		NIST Webbook
rinpol	1427.00		NIST Webbook
rinpol	1456.00		NIST Webbook

rinpol	1426.00	NIST Webbook
rinpol	1404.00	NIST Webbook
rinpol	1433.00	NIST Webbook
rinpol	1421.00	NIST Webbook
rinpol	1430.00	NIST Webbook
rinpol	1437.00	NIST Webbook
rinpol	1426.00	NIST Webbook
rinpol	1416.00	NIST Webbook
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ripol	1833.00		NIST Webbook
ripol	1857.00		NIST Webbook
ripol	1854.00		NIST Webbook
ripol	1825.00		NIST Webbook
ripol	1879.00		NIST Webbook
ripol	1798.00		NIST Webbook
tb	574.13	K	Joback Method
tc	792.17	K	Joback Method
tf	321.44	K	Joback Method
vc	0.665	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	438.46	J/mol×K	574.13	Joback Method
cpg	457.12	J/mol×K	610.47	Joback Method
cpg	474.66	J/mol×K	646.81	Joback Method
cpg	491.22	J/mol×K	683.15	Joback Method
cpg	506.92	J/mol×K	719.49	Joback Method
cpg	521.90	J/mol×K	755.83	Joback Method
cpg	536.27	J/mol×K	792.17	Joback Method

hvapt	62.00	kJ/mol	437.50	NIST Webbook
hvapt	67.50	kJ/mol	309.50	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	404.20	K	1.70	NIST Webbook

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C127413&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C127413&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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