

# 1,3,5-Trioxane, 2,4,6-tripropyl-

<b>Other names:</b>	s-Trioxane, 2,4,6-tripropyl- Parabutyraldehyde 2,4,6-Tripropyl-s-trioxane p-n-Butyraldehyde 2,4,6-tripropyl-1,3,5-trioxane
<b>Inchi:</b>	InChI=1S/C12H24O3/c1-4-7-10-13-11(8-5-2)15-12(14-10)9-6-3/h10-12H,4-9H2,1-3H3
<b>InchiKey:</b>	OXSRKPVVMMMER-UHFFFAOYSA-N
<b>Formula:</b>	C12H24O3
<b>SMILES:</b>	CCCC1OC(CCC)OC(CCC)O1
<b>Mol. weight [g/mol]:</b>	216.32
<b>CAS:</b>	2396-43-2

## Physical Properties

Property code	Value	Unit	Source
gf	-199.17	kJ/mol	Joback Method
hf	-673.37	kJ/mol	Joback Method
hfus	44.75	kJ/mol	Joback Method
hvap	55.65	kJ/mol	Joback Method
log10ws	-3.82		Crippen Method
logp	3.428		Crippen Method
mcvol	186.690	ml/mol	McGowan Method
pc	1961.34	kPa	Joback Method
tb	565.02	K	Joback Method
tc	754.58	K	Joback Method
tf	303.61	K	Joback Method
vc	0.702	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	503.09	J/molxK	565.02	Joback Method
cpg	591.43	J/molxK	722.98	Joback Method
cpg	575.53	J/molxK	691.39	Joback Method
cpg	558.75	J/molxK	659.80	Joback Method

cpg	541.09	J/molxK	628.21	Joback Method
cpg	522.54	J/molxK	596.61	Joback Method
cpg	606.48	J/molxK	754.58	Joback Method
dvisc	0.0002787	Paxs	565.02	Joback Method
dvisc	0.0003611	Paxs	521.45	Joback Method
dvisc	0.0004903	Paxs	477.88	Joback Method
dvisc	0.0007079	Paxs	434.31	Joback Method
dvisc	0.0011093	Paxs	390.75	Joback Method
dvisc	0.0019460	Paxs	347.18	Joback Method
dvisc	0.0040110	Paxs	303.61	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2396432&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2396432&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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