

# cis-11,14-Eicosadienoic acid, methyl ester

<b>Other names:</b>	(all-Z)-Methyl eicosa-11,14-dienoate 11,14-Eicosadienoic acid, methyl ester, (Z,Z) methyl Z,Z 11,14-eicosadienoate
<b>Inchi:</b>	InChI=1S/C21H38O2/c1-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-20-21(22)23-2/h7
<b>InchiKey:</b>	GWJCFAOQCNNFAM-NQLNTRKDSA-N
<b>Formula:</b>	C21H38O2
<b>SMILES:</b>	CCCCC=CCC=CCCCCCCCC(=O)OC
<b>Mol. weight [g/mol]:</b>	322.53

## Physical Properties

Property code	Value	Unit	Source
gf	52.46	kJ/mol	Joback Method
hf	-487.13	kJ/mol	Joback Method
hfus	53.34	kJ/mol	Joback Method
hvap	71.41	kJ/mol	Joback Method
log10ws	-7.18		Crippen Method
logp	6.753		Crippen Method
mcvol	305.590	ml/mol	McGowan Method
pc	1051.41	kPa	Joback Method
rinpol	2279.00		NIST Webbook
rinpol	2279.00		NIST Webbook
rinpol	2305.90		NIST Webbook
tb	764.49	K	Joback Method
tc	943.97	K	Joback Method
tf	388.43	K	Joback Method
vc	1.196	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	921.82	J/molxK	764.49	Joback Method
cpg	940.85	J/molxK	794.40	Joback Method
cpg	958.97	J/molxK	824.32	Joback Method
cpg	976.21	J/molxK	854.23	Joback Method

cpg	992.61	J/molxK	884.14	Joback Method
cpg	1008.23	J/molxK	914.06	Joback Method
cpg	1023.10	J/molxK	943.97	Joback Method
dvisc	0.0013095	Paxs	388.43	Joback Method
dvisc	0.0005151	Paxs	451.11	Joback Method
dvisc	0.0002544	Paxs	513.78	Joback Method
dvisc	0.0001465	Paxs	576.46	Joback Method
dvisc	0.0000940	Paxs	639.14	Joback Method
dvisc	0.0000653	Paxs	701.81	Joback Method
dvisc	0.0000481	Paxs	764.49	Joback Method
hvapt	117.50	kJ/mol	298.15	the vaporization enthalpies and vapor pressures of a series of unstaured fatty acid methyl esters by correlation gas chromatography

## Sources

NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U333618&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U333618&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
the vaporization enthalpies and vapor pressures of a series of unstaured fatty acid methyl esters by correlation gas chromatography:	<a href="https://www.doi.org/10.1016/j.tca.2007.02.008">https://www.doi.org/10.1016/j.tca.2007.02.008</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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