

# 4-Butylbenzotrile

<b>Other names:</b>	1-Butyl-4-cyanobenzene Benzonitrile,4-butyl-
<b>Inchi:</b>	InChI=1S/C11H13N/c1-2-3-4-10-5-7-11(9-12)8-6-10/h5-8H,2-4H2,1H3
<b>InchiKey:</b>	KGZNJCXNPLUEQS-UHFFFAOYSA-N
<b>Formula:</b>	C11H13N
<b>SMILES:</b>	CCCCc1ccc(C#N)cc1
<b>Mol. weight [g/mol]:</b>	159.23
<b>CAS:</b>	20651-73-4

## Physical Properties

Property code	Value	Unit	Source
gf	277.70	kJ/mol	Joback Method
hf	119.57	kJ/mol	Joback Method
hfus	19.40	kJ/mol	Joback Method
hvap	53.50	kJ/mol	Joback Method
ie	10.10 ± 0.10	eV	NIST Webbook
log10ws	-3.46		Crippen Method
logp	2.901		Crippen Method
mcvol	143.470	ml/mol	McGowan Method
pc	2543.05	kPa	Joback Method
tb	584.82	K	Joback Method
tc	805.11	K	Joback Method
tf	317.66	K	Joback Method
vc	0.570	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	333.73	J/molxK	584.82	Joback Method
cpg	346.84	J/molxK	621.53	Joback Method
cpg	359.15	J/molxK	658.25	Joback Method
cpg	370.71	J/molxK	694.96	Joback Method
cpg	381.54	J/molxK	731.68	Joback Method
cpg	391.67	J/molxK	768.39	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.34482e+01
Coeff. B	-4.04701e+03
Coeff. C	-8.95550e+01
Temperature range (K), min.	397.07
Temperature range (K), max.	586.93

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C20651734&Units=SI>

**The Yaws Handbook of Vapor Pressure:**

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307I>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure

**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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