

Tenamfetamine

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|-----------------------------|---|
| Other names: | 3,4-Methylenedioxy-amphetamine MDA Phenethylamine, «alpha»-methyl-3,4-(methylenedioxy)- 1,3-Benzodioxole-5-ethanamine, «alpha»-methyl- Methylenedioxyamphetamine «alpha»-Methyl-3,4-(methylenedioxy)phenethylamine (.+/-)-3,4-Methylenedioxyamphetamine |
| Inchi: | InChI=1S/C10H13NO2/c1-7(11)4-8-2-3-9-10(5-8)13-6-12-9/h2-3,5,7H,4,6,11H2,1H3 |
| InchiKey: | NGBBVGZWCFBOGO-UHFFFAOYSA-N |
| Formula: | C10H13NO2 |
| SMILES: | CC(N)Cc1ccc2c(c1)OCO2 |
| Mol. weight [g/mol]: | 179.22 |
| CAS: | 51497-09-7 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|-------------|---------|----------------|
| gf | 86.70 | kJ/mol | Joback Method |
| hf | -178.49 | kJ/mol | Joback Method |
| hfus | 29.62 | kJ/mol | Joback Method |
| hvap | 60.95 | kJ/mol | Joback Method |
| ie | 8.01 ± 0.06 | eV | NIST Webbook |
| log10ws | -2.45 | | Crippen Method |
| logp | 1.305 | | Crippen Method |
| mcvol | 138.860 | ml/mol | McGowan Method |
| pc | 3598.56 | kPa | Joback Method |
| tb | 602.24 | K | Joback Method |
| tc | 836.85 | K | Joback Method |
| tf | 397.50 | K | Joback Method |
| vc | 0.510 | m3/kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 357.51 | J/molxK | 602.24 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 371.07 | J/mol×K | 641.34 | Joback Method |
| cpg | 383.66 | J/mol×K | 680.44 | Joback Method |
| cpg | 395.34 | J/mol×K | 719.55 | Joback Method |
| cpg | 406.20 | J/mol×K | 758.65 | Joback Method |
| cpg | 416.30 | J/mol×K | 797.75 | Joback Method |
| cpg | 425.71 | J/mol×K | 836.85 | Joback Method |

Sources

| | |
|------------------------|---|
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=C51497097&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307I |

Legend

| | |
|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| ie: | Ionization energy |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| mcvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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