

# 7-Tetradecene

<b>Other names:</b>	7-Tetradecene,c&t Tetradec-7-ene
<b>Inchi:</b>	InChI=1S/C14H28/c1-3-5-7-9-11-13-14-12-10-8-6-4-2/h13-14H,3-12H2,1-2H3
<b>InchiKey:</b>	UBDIXSAEHLOROW-UHFFFAOYSA-N
<b>Formula:</b>	C14H28
<b>SMILES:</b>	CCCCCCC=CCCCCCC
<b>Mol. weight [g/mol]:</b>	196.37
<b>CAS:</b>	10374-74-0

## Physical Properties

Property code	Value	Unit	Source
gf	147.22	kJ/mol	Joback Method
hf	-215.07	kJ/mol	Joback Method
hfus	32.22	kJ/mol	Joback Method
hvap	46.72	kJ/mol	Joback Method
log10ws	-5.54		Crippen Method
logp	5.483		Crippen Method
mcvol	203.820	ml/mol	McGowan Method
pc	1587.28	kPa	Joback Method
rinpol	1367.00		NIST Webbook
rinpol	1367.00		NIST Webbook
rinpol	1370.00		NIST Webbook
tb	523.88	K	Joback Method
tc	689.82	K	Joback Method
tf	242.46	K	Joback Method
vc	0.799	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	488.90	J/molxK	523.88	Joback Method
cpg	506.77	J/molxK	551.54	Joback Method
cpg	523.90	J/molxK	579.19	Joback Method
cpg	540.31	J/molxK	606.85	Joback Method

cpg	556.03	J/mol×K	634.51	Joback Method
cpg	571.08	J/mol×K	662.16	Joback Method
cpg	585.50	J/mol×K	689.82	Joback Method
dvisc	0.0054364	Paxs	242.46	Joback Method
dvisc	0.0018593	Paxs	289.36	Joback Method
dvisc	0.0008577	Paxs	336.27	Joback Method
dvisc	0.0004782	Paxs	383.17	Joback Method
dvisc	0.0003029	Paxs	430.07	Joback Method
dvisc	0.0002098	Paxs	476.98	Joback Method
dvisc	0.0001553	Paxs	523.88	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.73911e+01
Coeff. B	-5.44627e+03
Coeff. C	-9.00450e+01
Temperature range (K), min.	408.48
Temperature range (K), max.	540.91

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10374740&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10374740&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/70-229-0/7-Tetradecene.pdf>

Generated by Cheméo on 2024-04-25 04:26:56.881963241 +0000 UTC m=+16308465.802540564.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.