

# Oxacyclotetradecan-2-one

<b>Other names:</b>	Tridecanoic acid, 13-hydroxy-, «mu»-lactone Tridecanolide 1,13-Tridecanolide 1-Oxacyclotetradecan-2-one 13-Tridecanolide Tridecano-13-lactone
<b>Inchi:</b>	InChI=1S/C13H24O2/c14-13-11-9-7-5-3-1-2-4-6-8-10-12-15-13/h1-12H2
<b>InchiKey:</b>	IUDIJIVSWGJNV-UHFFFAOYSA-N
<b>Formula:</b>	C13H24O2
<b>SMILES:</b>	O=C1CCCCCCCCCCCCO1
<b>Mol. weight [g/mol]:</b>	212.33
<b>CAS:</b>	1725-04-8

## Physical Properties

Property code	Value	Unit	Source
chs	-7895.30 ± 2.90	kJ/mol	NIST Webbook
gf	-214.77	kJ/mol	Joback Method
hf	-569.00 ± 3.00	kJ/mol	NIST Webbook
hfl	-644.00 ± 2.00	kJ/mol	NIST Webbook
hfs	-648.10 ± 2.90	kJ/mol	NIST Webbook
hfus	10.88	kJ/mol	Joback Method
hvap	72.90 ± 1.70	kJ/mol	NIST Webbook
hvap	75.00	kJ/mol	NIST Webbook
log10ws	-4.02		Crippen Method
logp	3.834		Crippen Method
mcvol	190.610	ml/mol	McGowan Method
pc	2495.01	kPa	Joback Method
rinpol	1590.00		NIST Webbook
rinpol	1636.00		NIST Webbook
rinpol	1627.00		NIST Webbook
rinpol	1640.80		NIST Webbook
rinpol	1627.00		NIST Webbook
rinpol	1636.00		NIST Webbook
ripol	2082.00		NIST Webbook
ripol	2082.00		NIST Webbook
ripol	2062.00		NIST Webbook
ripol	2055.00		NIST Webbook

ripol	2030.00		NIST Webbook
ripol	2030.00		NIST Webbook
ss	401.90	J/molxK	NIST Webbook
ss	401.80	J/molxK	NIST Webbook
tb	649.99	K	Joback Method
tc	913.37	K	Joback Method
tf	314.52	K	Joback Method
tt	300.42 ± 0.01	K	NIST Webbook
tt	300.40 ± 0.10	K	NIST Webbook
vc	0.661	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	620.58	J/molxK	781.68	Joback Method
cpg	641.85	J/molxK	825.58	Joback Method
cpg	660.68	J/molxK	869.48	Joback Method
cpg	542.95	J/molxK	649.99	Joback Method
cpg	571.03	J/molxK	693.89	Joback Method
cpg	596.94	J/molxK	737.78	Joback Method
cpg	676.97	J/molxK	913.37	Joback Method
cps	398.30	J/molxK	298.15	NIST Webbook
cps	398.30	J/molxK	298.15	NIST Webbook
hfust	9.08	kJ/mol	300.40	NIST Webbook
hvapt	66.60 ± 1.10	kJ/mol	390.00	NIST Webbook
hvapt	67.50	kJ/mol	418.00	NIST Webbook

## Sources

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1725048&Units=SI>

# Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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