

# Dipropylene glycol, # 1

<b>Inchi:</b>	InChI=1S/C6H14O3/c1-5(7)3-9-4-6(2)8/h5-8H,3-4H2,1-2H3
<b>InchiKey:</b>	AZUXKVXMJOIAOF-UHFFFAOYSA-N
<b>Formula:</b>	C6H14O3
<b>SMILES:</b>	CC(O)COCC(C)O
<b>Mol. weight [g/mol]:</b>	134.17

## Physical Properties

Property code	Value	Unit	Source
gf	-383.88	kJ/mol	Joback Method
hf	-614.41	kJ/mol	Joback Method
hfus	13.61	kJ/mol	Joback Method
hvap	63.94	kJ/mol	Joback Method
log10ws	-0.17		Crippen Method
logp	-0.235		Crippen Method
mcvol	113.010	ml/mol	McGowan Method
pc	3853.09	kPa	Joback Method
rinpol	1017.00		NIST Webbook
ripol	1767.00		NIST Webbook
ripol	1767.00		NIST Webbook
tb	542.58	K	Joback Method
tc	705.59	K	Joback Method
tf	271.25	K	Joback Method
vc	0.415	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	278.16	J/mol×K	542.58	Joback Method
cpg	319.10	J/mol×K	678.42	Joback Method
cpg	311.54	J/mol×K	651.25	Joback Method
cpg	303.67	J/mol×K	624.08	Joback Method
cpg	295.49	J/mol×K	596.92	Joback Method
cpg	286.98	J/mol×K	569.75	Joback Method
cpg	326.35	J/mol×K	705.59	Joback Method

dvisc	0.0000478	Paxs	542.58	Joback Method
dvisc	0.0001018	Paxs	497.36	Joback Method
dvisc	0.0002523	Paxs	452.14	Joback Method
dvisc	0.0007649	Paxs	406.91	Joback Method
dvisc	0.0030600	Paxs	361.69	Joback Method
dvisc	0.0181938	Paxs	316.47	Joback Method
dvisc	0.1960058	Paxs	271.25	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R409666&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R409666&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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