

# CH<sub>3</sub>C(O)CH<sub>2</sub>CH<sub>2</sub>OH

<b>Other names:</b>	Methylolacetone Monomethylolacetone 3-Ketobutan-1-ol 3-Oxo-1-butanol 3-Oxobutanol 4-Butanol-2-one 4-Hydroxy-2-butanone NSC 41219 4-hydroxybutan-2-one
<b>Inchi:</b>	InChI=1S/C4H8O2/c1-4(6)2-3-5/h5H,2-3H2,1H3
<b>InchiKey:</b>	LVSQXDHWDCMMRJ-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>4</sub> H <sub>8</sub> O <sub>2</sub>
<b>SMILES:</b>	CC(=O)CCO
<b>Mol. weight [g/mol]:</b>	88.11
<b>CAS:</b>	590-90-9

## Physical Properties

Property code	Value	Unit	Source
gf	-282.94	kJ/mol	Joback Method
hf	-390.70	kJ/mol	Joback Method
hfus	11.80	kJ/mol	Joback Method
hvap	47.92	kJ/mol	Joback Method
log10ws	-0.04		Crippen Method
logp	-0.042		Crippen Method
mcvol	74.660	ml/mol	McGowan Method
pc	4775.98	kPa	Joback Method
rinpol	755.00		NIST Webbook
rinpol	814.00		NIST Webbook
rinpol	751.00		NIST Webbook
rinpol	798.00		NIST Webbook
rinpol	798.00		NIST Webbook
rinpol	751.00		NIST Webbook
tb	436.97	K	Joback Method
tc	610.84	K	Joback Method
tf	245.59	K	Joback Method
vc	0.284	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	146.10	J/molxK	436.97	Joback Method
cpg	152.68	J/molxK	465.95	Joback Method
cpg	159.00	J/molxK	494.93	Joback Method
cpg	165.07	J/molxK	523.90	Joback Method
cpg	170.89	J/molxK	552.88	Joback Method
cpg	176.47	J/molxK	581.86	Joback Method
cpg	181.81	J/molxK	610.84	Joback Method
dvisc	0.0318897	Paxs	245.59	Joback Method
dvisc	0.0092130	Paxs	277.49	Joback Method
dvisc	0.0034383	Paxs	309.38	Joback Method
dvisc	0.0015427	Paxs	341.28	Joback Method
dvisc	0.0007939	Paxs	373.18	Joback Method
dvisc	0.0004536	Paxs	405.07	Joback Method
dvisc	0.0002812	Paxs	436.97	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	382.70	K	4.00	NIST Webbook
tbrp	363.20	K	1.50	NIST Webbook

# Sources

- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)
- Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)
- McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C590909&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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