

# Norbornane, 2-bromo-3-(trichloromethyl), exo-Br

Inchi:	InChI=1S/C8H10BrCl3/c9-7-5-2-1-4(3-5)6(7)8(10,11)12/h4-7H,1-3H2/t4?,5?,6-,7+/m0/s1
InchiKey:	JBMOBFXXJVIZLP-GOXNJFHPSA-N
Formula:	C8H10BrCl3
SMILES:	C1C(Cl)(Cl)C1C2CCC(C2)C1Br
Mol. weight [g/mol]:	292.43

## Physical Properties

Property code	Value	Unit	Source
gf	91.83	kJ/mol	Joback Method
hf	-139.33	kJ/mol	Joback Method
hfus	23.25	kJ/mol	Joback Method
hvap	51.08	kJ/mol	Joback Method
log10ws	-4.34		Crippen Method
logp	4.166		Crippen Method
mcvol	156.080	ml/mol	McGowan Method
pc	3121.00	kPa	Joback Method
rinsol	1540.00		NIST Webbook
tb	566.07	K	Joback Method
tc	817.12	K	Joback Method
tf	355.78	K	Joback Method
vc	0.586	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	336.05	J/molxK	566.07	Joback Method
cpg	350.91	J/molxK	607.91	Joback Method
cpg	364.37	J/molxK	649.75	Joback Method
cpg	376.58	J/molxK	691.60	Joback Method
cpg	387.69	J/molxK	733.44	Joback Method
cpg	397.84	J/molxK	775.28	Joback Method
cpg	407.17	J/molxK	817.12	Joback Method
dvisc	0.0028279	Paxs	355.78	Joback Method
dvisc	0.0023177	Paxs	390.83	Joback Method

dvisc	0.0019628	Paxs	425.88	Joback Method
dvisc	0.0017048	Paxs	460.93	Joback Method
dvisc	0.0015105	Paxs	495.97	Joback Method
dvisc	0.0013599	Paxs	531.02	Joback Method
dvisc	0.0012403	Paxs	566.07	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R515385&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R515385&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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