

# trichlorotoluene

<b>Other names:</b>	Benzene, trichloromethyl-
<b>Inchi:</b>	InChI=1S/C7H5Cl3/c8-7(9,10)6-4-2-1-3-5-6/h1-5H
<b>InchiKey:</b>	XEMRAKSQROQPBR-UHFFFAOYSA-N
<b>Formula:</b>	C7H5Cl3
<b>SMILES:</b>	C1C(Cl)(Cl)C(Cl)C=C=C1
<b>Mol. weight [g/mol]:</b>	195.47
<b>CAS:</b>	30583-33-6

## Physical Properties

Property code	Value	Unit	Source
chl	-3494.90 ± 1.80	kJ/mol	NIST Webbook
gf	87.52	kJ/mol	Joback Method
hf	12.50 ± 2.30	kJ/mol	NIST Webbook
hfl	-45.10 ± 2.00	kJ/mol	NIST Webbook
hfus	13.10	kJ/mol	Joback Method
hvap	57.60 ± 1.20	kJ/mol	NIST Webbook
log10ws	-3.47		Crippen Method
logp	3.513		Crippen Method
mvol	122.450	ml/mol	McGowan Method
pc	3668.65	kPa	Joback Method
tb	495.30	K	Joback Method
tc	744.27	K	Joback Method
tf	287.25	K	Joback Method
vc	0.456	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	212.89	J/mol×K	495.30	Joback Method
cpg	223.35	J/mol×K	536.79	Joback Method
cpg	232.76	J/mol×K	578.29	Joback Method
cpg	241.20	J/mol×K	619.78	Joback Method
cpg	248.77	J/mol×K	661.28	Joback Method
cpg	255.53	J/mol×K	702.77	Joback Method

cpg	261.58	J/mol×K	744.27	Joback Method
dvisc	0.0039216	Paxs	287.25	Joback Method
dvisc	0.0020289	Paxs	321.93	Joback Method
dvisc	0.0011932	Paxs	356.60	Joback Method
dvisc	0.0007709	Paxs	391.27	Joback Method
dvisc	0.0005348	Paxs	425.95	Joback Method
dvisc	0.0003920	Paxs	460.62	Joback Method
dvisc	0.0003001	Paxs	495.30	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C30583336&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C30583336&amp;Units=SI</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/70-463-0/trichlorotoluene.pdf>

Generated by Cheméo on 2024-04-28 15:37:38.769510924 +0000 UTC m=+16607907.690088256.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.