

# 2-Pentenenitrile

<b>Other names:</b>	1-Cyano-1-butene 2-Pentenitrile Pent-2-enitrile
<b>Inchi:</b>	InChI=1S/C5H7N/c1-2-3-4-5-6/h3-4H,2H2,1H3/b4-3+
<b>InchiKey:</b>	ISBHMJZRKAFTGE-ONEGZZNKSA-N
<b>Formula:</b>	C5H7N
<b>SMILES:</b>	CCC=CC#N
<b>Mol. weight [g/mol]:</b>	81.12
<b>CAS:</b>	13284-42-9

## Physical Properties

Property code	Value	Unit	Source
gf	204.62	kJ/mol	Joback Method
hf	135.57	kJ/mol	Joback Method
hfus	10.41	kJ/mol	Joback Method
hvap	37.16	kJ/mol	Joback Method
log10ws	-1.64		Crippen Method
logp	1.476		Crippen Method
mcvol	78.390	ml/mol	McGowan Method
pc	3581.35	kPa	Joback Method
rinsol	755.00		NIST Webbook
tb	420.04	K	Joback Method
tc	620.58	K	Joback Method
tf	206.02	K	Joback Method
vc	0.322	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	136.82	J/mol×K	420.04	Joback Method
cpg	144.23	J/mol×K	453.46	Joback Method
cpg	151.25	J/mol×K	486.89	Joback Method
cpg	157.88	J/mol×K	520.31	Joback Method
cpg	164.16	J/mol×K	553.73	Joback Method

cpg	170.09	J/mol×K	587.15	Joback Method
cpg	175.70	J/mol×K	620.58	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C13284429&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C13284429&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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