

# 1,2-Bis(tert-butylimino)ethane

<b>Other names:</b>	1,2-Di(N-tert-butylimino)ethane 1,4-Di-tert-butyl-1,4-diazabutadiene 1,4-Diazabutadiene, 1,4-di-tert-butyl- 2-Propanamine, N,N'-1,2-ethanediylidenebis[2-methyl- Bis(tert-butylimino)ethane Ethylamine, N,N'-ethanediylidenebis[1,1-dimethyl- N,N'-Di-tert-butylethanediimine N-(2-([1,1-Dimethylethyl]imino)ethylidene)-2-methyl-2-propanamine
<b>Inchi:</b>	InChI=1S/C10H20N2/c1-9(2,3)11-7-8-12-10(4,5)6/h7-8H,1-6H3
<b>InchiKey:</b>	HACCVLBYBQLWMC-UHFFFAOYSA-N
<b>Formula:</b>	C10H20N2
<b>SMILES:</b>	CC(C)(C)N=CC=NC(C)(C)C
<b>Mol. weight [g/mol]:</b>	168.28
<b>CAS:</b>	30834-74-3

## Physical Properties

Property code	Value	Unit	Source
hf	-102.79	kJ/mol	Joback Method
hvap	41.89	kJ/mol	Joback Method
ie	8.50	eV	NIST Webbook
ie	9.26	eV	NIST Webbook
log10ws	-2.56		Crippen Method
logp	2.725		Crippen Method
mcvol	163.120	ml/mol	McGowan Method
pc	1810.77	kPa	Joback Method
tb	575.10	K	Joback Method
tc	796.40	K	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C24764883&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C24764883&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307i">http://pubs.acs.org/doi/abs/10.1021/ci990307i</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logP:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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