

# Lilac alcohol B

<b>Other names:</b>	2-Furanethanol, tetrahydro-«beta»,5-dimethyl-5-vinyl-, («beta»R,2S,5S)- Lilac alcohol (isomer II)
<b>Inchi:</b>	InChI=1S/C10H18O2/c1-4-10(3)6-5-9(12-10)8(2)7-11/h4,8-9,11H,1,5-7H2,2-3H3
<b>InchiKey:</b>	VUEGXHXUMOZKKN-UHFFFAOYSA-N
<b>Formula:</b>	C10H18O2
<b>SMILES:</b>	C=CC1(C)CCC(C(C)CO)O1
<b>Mol. weight [g/mol]:</b>	170.25
<b>CAS:</b>	33081-35-5

## Physical Properties

Property code	Value	Unit	Source
gf	-80.87	kJ/mol	Joback Method
hf	-358.43	kJ/mol	Joback Method
hfus	17.63	kJ/mol	Joback Method
hvap	56.78	kJ/mol	Joback Method
log10ws	-2.09		Crippen Method
logp	1.739		Crippen Method
mcvol	148.340	ml/mol	McGowan Method
pc	2912.39	kPa	Joback Method
rinpol	1212.00		NIST Webbook
rinpol	1219.00		NIST Webbook
rinpol	1219.00		NIST Webbook
rinpol	1219.00		NIST Webbook
rinpol	1219.00		NIST Webbook
ripol	1756.00		NIST Webbook
ripol	1758.00		NIST Webbook
tb	554.42	K	Joback Method
tc	747.74	K	Joback Method
tf	303.65	K	Joback Method
vc	0.548	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	380.87	J/mol×K	554.42	Joback Method
cpg	395.72	J/mol×K	586.64	Joback Method
cpg	409.73	J/mol×K	618.86	Joback Method
cpg	422.99	J/mol×K	651.08	Joback Method
cpg	435.60	J/mol×K	683.30	Joback Method
cpg	447.63	J/mol×K	715.52	Joback Method
cpg	459.18	J/mol×K	747.74	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C33081355&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C33081355&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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