

# Bicyclo[3.1.0]hexan-2-ol, 2-methyl-5-(1-methylethyl)-

Other names:	4-Thujanol Sabinene hydrate (Z)-Sabinene hydrate 2-methyl-5-(1-methylethyl)bicyclo[3.1.0]hexan-2-ol
Inchi:	InChI=1S/C10H18O/c1-7(2)10-5-4-9(3,11)8(10)6-10/h7-8,11H,4-6H2,1-3H3
InchiKey:	KXSDPILWMGFJMM-UHFFFAOYSA-N
Formula:	C10H18O
SMILES:	CC(C)C12CCC(C)(O)C1C2
Mol. weight [g/mol]:	154.25
CAS:	546-79-2

## Physical Properties

Property code	Value	Unit	Source
gf	-3.13	kJ/mol	Joback Method
hf	-251.50	kJ/mol	Joback Method
hfus	6.97	kJ/mol	Joback Method
hvap	51.36	kJ/mol	Joback Method
log10ws	-2.45		Crippen Method
logp	2.194		Crippen Method
mcvol	135.910	ml/mol	McGowan Method
pc	3224.64	kPa	Joback Method
rinpol	1053.00		NIST Webbook
rinpol	1051.00		NIST Webbook
rinpol	1082.00		NIST Webbook
rinpol	1068.00		NIST Webbook
rinpol	1109.00		NIST Webbook
rinpol	1095.00		NIST Webbook
rinpol	1090.00		NIST Webbook
rinpol	1033.00		NIST Webbook
rinpol	1068.00		NIST Webbook
rinpol	1070.00		NIST Webbook
rinpol	1094.00		NIST Webbook
rinpol	1096.00		NIST Webbook
rinpol	1070.00		NIST Webbook
rinpol	1102.00		NIST Webbook
rinpol	1096.00		NIST Webbook
rinpol	1077.00		NIST Webbook

rinpol	1082.00		NIST Webbook
rinpol	1040.00		NIST Webbook
rinpol	1109.00		NIST Webbook
rinpol	1070.00		NIST Webbook
rinpol	1051.00		NIST Webbook
rinpol	1114.00		NIST Webbook
rinpol	1068.00		NIST Webbook
rinpol	1040.00		NIST Webbook
rinpol	1125.00		NIST Webbook
ripol	1469.00		NIST Webbook
ripol	1469.00		NIST Webbook
tb	529.23	K	Joback Method
tc	727.62	K	Joback Method
tf	327.72	K	Joback Method
vc	0.517	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	353.14	J/mol×K	529.23	Joback Method
cpg	368.19	J/mol×K	562.29	Joback Method
cpg	382.12	J/mol×K	595.36	Joback Method
cpg	395.11	J/mol×K	628.42	Joback Method
cpg	407.37	J/mol×K	661.49	Joback Method
cpg	419.07	J/mol×K	694.55	Joback Method
cpg	430.42	J/mol×K	727.62	Joback Method

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C546792&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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