

# Mandelic acid

<b>Other names:</b>	Benzeneacetic acid, «alpha»-hydroxy- «alpha»-Hydroxy-«alpha»-toluic acid «alpha»-Hydroxyphenylacetic acid p-Mandelic acid Almond acid Amygdalic acid Amygdalinic acid Phenylglycolic acid Phenylhydroxyacetic acid Uromaline dl-Mandelic acid «alpha»-Toluic acid, «alpha»-hydroxy- Glycolic acid, phenyl- Racemic mandelic acid Kyselina 2-fenyl-2-hydroxyethanova Kyselina mandlova Paramandelic acid «alpha»-Phenylhydroxyacetic acid Benzoglycolic acid Hydroxy(phenyl)acetic acid Benzeneacetic acid, «alpha»-hydroxy-, (.+/-)- (.+/-)-alpha-Hydroxybenzeneacetic acid (.+/-)-Mandelic acid (RS)-Mandelic acid 2-Phenyl-2-hydroxyacetic acid 2-Phenylglycolic acid DL-Hydroxy(phenyl)acetic acid NSC 7925 114-21-6 530-31-4
<b>Inchi:</b>	InChI=1S/C8H8O3/c9-7(8(10)11)6-4-2-1-3-5-6/h1-5,7,9H,(H,10,11)
<b>InchiKey:</b>	IWYDHOAUDWTVEP-UHFFFAOYSA-N
<b>Formula:</b>	C8H8O3
<b>SMILES:</b>	O=C(O)C(O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	152.15
<b>CAS:</b>	90-64-2

# Physical Properties

Property code	Value	Unit	Source
gf	-276.11	kJ/mol	Joback Method
hf	-394.24	kJ/mol	Joback Method
hfus	16.77	kJ/mol	Joback Method
hvap	75.39	kJ/mol	Joback Method
log10ws	-1.10		Crippen Method
logp	0.805		Crippen Method
mcvol	113.130	ml/mol	McGowan Method
pc	5102.04	kPa	Joback Method
rinpol	1487.00		NIST Webbook
tb	646.91	K	Joback Method
tc	844.13	K	Joback Method
tf	392.00 ± 1.50	K	NIST Webbook
vc	0.413	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	313.57	J/molxK	811.26	Joback Method
cpg	278.92	J/molxK	646.91	Joback Method
cpg	286.85	J/molxK	679.78	Joback Method
cpg	294.26	J/molxK	712.65	Joback Method
cpg	301.16	J/molxK	745.52	Joback Method
cpg	307.59	J/molxK	778.39	Joback Method
cpg	319.12	J/molxK	844.13	Joback Method
cps	199.20	J/molxK	323.00	NIST Webbook
dvisc	0.0000249	Paxs	646.91	Joback Method
dvisc	0.0084829	Paxs	362.91	Joback Method
dvisc	0.0018322	Paxs	410.24	Joback Method
dvisc	0.0005434	Paxs	457.58	Joback Method
dvisc	0.0002024	Paxs	504.91	Joback Method
dvisc	0.0000893	Paxs	552.24	Joback Method
dvisc	0.0000448	Paxs	599.58	Joback Method
hfust	25.52	kJ/mol	392.00	NIST Webbook

# Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C90642&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C90642&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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