

# O-sec-amylphenol

<b>Other names:</b>	2-(1-methylbutyl)phenol Phenol, 2-(1-methylbutyl)-
<b>Inchi:</b>	InChI=1S/C11H16O/c1-3-6-9(2)10-7-4-5-8-11(10)12/h4-5,7-9,12H,3,6H2,1-2H3
<b>InchiKey:</b>	ROMXEVFSCNLHAB-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O
<b>SMILES:</b>	CCCC(C)c1ccccc1O
<b>Mol. weight [g/mol]:</b>	164.24
<b>CAS:</b>	87-26-3

## Physical Properties

Property code	Value	Unit	Source
gf	-2.91	kJ/mol	Joback Method
hf	-216.43	kJ/mol	Joback Method
hfus	20.55	kJ/mol	Joback Method
hvap	54.98	kJ/mol	Joback Method
log10ws	-3.04		Crippen Method
logp	3.296		Crippen Method
mvol	147.960	ml/mol	McGowan Method
pc	3135.00	kPa	Joback Method
tb	515.74 ± 0.01	K	NIST Webbook
tc	777.32	K	Joback Method
tf	336.87	K	Joback Method
vc	0.503	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	423.84	J/mol×K	740.76	Joback Method
cpg	412.32	J/mol×K	704.20	Joback Method
cpg	400.05	J/mol×K	667.63	Joback Method
cpg	386.96	J/mol×K	631.07	Joback Method
cpg	372.96	J/mol×K	594.50	Joback Method
cpg	357.97	J/mol×K	557.94	Joback Method
cpg	434.70	J/mol×K	777.32	Joback Method

dvisc	0.0046044	Paxs	336.87	Joback Method
dvisc	0.0000452	Paxs	557.94	Joback Method
dvisc	0.0000744	Paxs	521.10	Joback Method
dvisc	0.0001322	Paxs	484.25	Joback Method
dvisc	0.0002578	Paxs	447.41	Joback Method
dvisc	0.0005672	Paxs	410.56	Joback Method
dvisc	0.0014575	Paxs	373.72	Joback Method
hvapt	59.60	kJ/mol	449.00	NIST Webbook
hvapt	74.40	kJ/mol	449.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.54521e+01
Coeff. B	-4.67166e+03
Coeff. C	-8.45260e+01
Temperature range (K), min.	392.59
Temperature range (K), max.	545.22

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C87263&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C87263&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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