

# Benzene, 1,4-diiodo-

<b>Other names:</b>	1,4-Diiodbenzene 1,4-Diiodobenzene Benzene, 1,4-iodo- Benzene, p-diiodo- p-Benzene diiodide p-Diiodobenzene
<b>Inchi:</b>	InChI=1S/C6H4I2/c7-5-1-2-6(8)4-3-5/h1-4H
<b>InchiKey:</b>	LFMWZTSOMGDDJU-UHFFFAOYSA-N
<b>Formula:</b>	C6H4I2
<b>SMILES:</b>	Ic1ccc(I)cc1
<b>Mol. weight [g/mol]:</b>	329.90
<b>CAS:</b>	624-38-4

## Physical Properties

Property code	Value	Unit	Source
chs	-3093.20 ± 4.20	kJ/mol	NIST Webbook
gf	218.66	kJ/mol	Joback Method
hf	211.63	kJ/mol	Joback Method
hfus	13.76	kJ/mol	Joback Method
hvap	50.63	kJ/mol	Joback Method
ie	8.60	eV	NIST Webbook
ie	8.45	eV	NIST Webbook
log10ws	-5.37		Aqueous Solubility Prediction Method
logp	2.896		Crippen Method
mcvol	123.280	ml/mol	McGowan Method
pc	4316.89	kPa	Joback Method
rinpol	1412.00		NIST Webbook
rinpol	1412.00		NIST Webbook
rinpol	1448.00		NIST Webbook
rinpol	1448.00		NIST Webbook
rinpol	1450.00		NIST Webbook
ripol	2176.00		NIST Webbook
ripol	2176.00		NIST Webbook
ripol	2176.00		NIST Webbook
tb	558.20	K	NIST Webbook
tc	852.28	K	Joback Method

tf	403.15	K	Thermal, physicochemical and microstructural studies of organic analog of nonmetal-nonmetal monotectic alloy
tf	402.00 ± 2.00	K	NIST Webbook
tf	400.00 ± 2.00	K	NIST Webbook
tf	402.15 ± 0.50	K	NIST Webbook
tf	403.98	K	Aqueous Solubility Prediction Method
vc	0.440	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	213.79	J/molxK	852.28	Joback Method
cpg	192.55	J/molxK	653.84	Joback Method
cpg	198.74	J/molxK	703.45	Joback Method
cpg	204.27	J/molxK	753.06	Joback Method
cpg	209.25	J/molxK	802.67	Joback Method
cpg	177.80	J/molxK	554.62	Joback Method
cpg	185.61	J/molxK	604.23	Joback Method
cps	160.77	J/molxK	298.15	NIST Webbook
dvisc	0.0005391	Paxs	473.89	Joback Method
dvisc	0.0007337	Paxs	433.53	Joback Method
dvisc	0.0010637	Paxs	393.17	Joback Method
dvisc	0.0016788	Paxs	352.80	Joback Method
dvisc	0.0003330	Paxs	554.62	Joback Method
dvisc	0.0004158	Paxs	514.26	Joback Method
dvisc	0.0029814	Paxs	312.44	Joback Method
hfust	22.37	kJ/mol	402.00	NIST Webbook
hfust	22.37	kJ/mol	402.00	NIST Webbook
hfust	22.30	kJ/mol	402.40	NIST Webbook
hfust	22.38	kJ/mol	402.15	NIST Webbook
hfust	22.34	kJ/mol	402.00	NIST Webbook
hsubt	63.40	kJ/mol	386.50	NIST Webbook
hvapt	52.60	kJ/mol	481.00	NIST Webbook
sfust	55.60	J/molxK	402.00	NIST Webbook
sfust	55.60	J/molxK	402.15	NIST Webbook
sfust	55.65	J/molxK	402.00	NIST Webbook

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51398e+01
Coeff. B	-4.85971e+03
Coeff. C	-9.63160e+01
Temperature range (K), min.	423.52
Temperature range (K), max.	590.77

## Sources

<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Thermal, physicochemical and microstructural studies of organic and inorganic metal-nonmetal monotectic alloy:</b>	<a href="https://www.doi.org/10.1016/j.tca.2009.10.010">https://www.doi.org/10.1016/j.tca.2009.10.010</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C624384&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C624384&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l

<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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