

# 2-Cyclohexen-1-ol, 2-methyl-5-(1-methylethenyl)-, propanoate

Other names:

p-Mentha-6,8-dien-2-ol, propionate

Carveol propionate

Carvyl propionate

l-Carvyl propionate

2-Cyclohexen-1-ol, 2-methyl-5-(1-methylethenyl)-, propionate

l-p-Mentha-6,8(9)-dien-2-yl propionate

p-mentha-1(6),8-dien-2-yl propionate

**Inchi:** InChI=1S/C13H20O2/c1-5-13(14)15-12-8-11(9(2)3)7-6-10(12)4/h6,11-12H,2,5,7-8H2,1,3

**InchiKey:** DFVXNZOMAOGTBL-UHFFFAOYSA-N

**Formula:** C13H20O2

**SMILES:** C=C(C)C1CC=C(C)C(OC(=O)CC)C1

**Mol. weight [g/mol]:** 208.30

**CAS:** 97-45-0

## Physical Properties

Property code	Value	Unit	Source
gf	-58.98	kJ/mol	Joback Method
hf	-360.52	kJ/mol	Joback Method
hfus	23.36	kJ/mol	Joback Method
hvap	54.17	kJ/mol	Joback Method
log10ws	-3.60		Crippen Method
logp	3.241		Crippen Method
mcvol	182.010	ml/mol	McGowan Method
pc	2104.20	kPa	Joback Method
rinpol	1441.00		NIST Webbook
rinpol	1440.00		NIST Webbook
rinpol	1440.00		NIST Webbook
rinpol	1440.00		NIST Webbook
rinpol	1441.00		NIST Webbook
ripol	1833.00		NIST Webbook
ripol	1833.00		NIST Webbook
ripol	1833.00		NIST Webbook
tb	588.71	K	Joback Method
tc	795.23	K	Joback Method
tf	309.13	K	Joback Method
vc	0.688	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	466.04	J/mol×K	588.71	Joback Method
cpg	484.54	J/mol×K	623.13	Joback Method
cpg	502.05	J/mol×K	657.55	Joback Method
cpg	518.61	J/mol×K	691.97	Joback Method
cpg	534.21	J/mol×K	726.39	Joback Method
cpg	548.87	J/mol×K	760.81	Joback Method
cpg	562.62	J/mol×K	795.23	Joback Method

# Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	352.20	K	0.03	NIST Webbook

## Sources

McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C97450&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C97450&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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