

# 1-Pentene, 4-chloro-

<b>Other names:</b>	4-Chloro-1-pentene
<b>Inchi:</b>	InChI=1S/C5H9Cl/c1-3-4-5(2)6/h3,5H,1,4H2,2H3
<b>InchiKey:</b>	BIGJOILKRSNJMQ-UHFFFAOYSA-N
<b>Formula:</b>	C5H9Cl
<b>SMILES:</b>	C=CCC(C)Cl
<b>Mol. weight [g/mol]:</b>	104.58
<b>CAS:</b>	10524-08-0

## Physical Properties

Property code	Value	Unit	Source
gf	64.69	kJ/mol	Joback Method
hf	-42.12	kJ/mol	Joback Method
hfus	8.10	kJ/mol	Joback Method
hvap	30.05	kJ/mol	Joback Method
log10ws	-2.03		Crippen Method
logp	2.190		Crippen Method
mcvol	89.250	ml/mol	McGowan Method
pc	3522.09	kPa	Joback Method
rinpol	700.00		NIST Webbook
tb	347.47	K	Joback Method
tc	528.00	K	Joback Method
tf	159.27	K	Joback Method
vc	0.340	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	137.93	J/molxK	347.47	Joback Method
cpg	146.57	J/molxK	377.56	Joback Method
cpg	154.83	J/molxK	407.65	Joback Method
cpg	162.72	J/molxK	437.74	Joback Method
cpg	170.25	J/molxK	467.83	Joback Method
cpg	177.44	J/molxK	497.91	Joback Method
cpg	184.29	J/molxK	528.00	Joback Method

dvisc	0.0062892	Paxs	159.27	Joback Method
dvisc	0.0024182	Paxs	190.64	Joback Method
dvisc	0.0012181	Paxs	222.00	Joback Method
dvisc	0.0007271	Paxs	253.37	Joback Method
dvisc	0.0004863	Paxs	284.74	Joback Method
dvisc	0.0003523	Paxs	316.10	Joback Method
dvisc	0.0002705	Paxs	347.47	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.43048e+01
Coeff. B	-3.22492e+03
Coeff. C	-4.40520e+01
Temperature range (K), min.	274.12
Temperature range (K), max.	402.64

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C10524080&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C10524080&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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