

# Pentaerythritol tetrakis[3-(3',5'-di-tert-butyl-4'-hydroxyphenyl)propanoate]

Other names:

ADK Stab AO 60

AO 60

Anox 20

Anox 20AM

BP 101

Benzenepropanoic acid, 3,5-bis(1,1-dimethylethyl)-4-hydroxy-,

1,1'-[2,2-bis[[3-[3,5-bis(1,1-dimethylethyl)-4-hydroxyphenyl]-1-oxopropoxy]methyl]-1,3-propanediyl]diester

Benzenepropanoic acid, 3,5-bis(1,1-dimethylethyl)-4-hydroxy-,

2,2-bis[[3-[3,5-bis(1,1-dimethylethyl)-4-hydroxyphenyl]-1-oxopropoxy]methyl]-1,3-propanediyl ester

Dovernox 10

Fenozan 22

Fenozan 23

Hydrocinnamic acid, 3,5-di-tert-butyl-4-hydroxy-, neopentanetetrayl ester

Hydrocinnamic acid, 3,5-di-tert-butyl-4-hydroxy-, tetraester with pentaerythritol

IR 1010

Irganox 1010FF

Irganox 1040

Lowinox PP35

MARK AO 60

Naugard 10

Neopentanetetrayl 3,5-di-tert-butyl-4-hydroxyhydrocinnamate

Pentaerythritol, tetrakis(3,5-di-tert-butyl-4-hydroxyhydrocinnamate)

Pentaerythrityl tetrakis-3-(3',5'-di-t-butyl-4'-hydroxyphenyl)propionate

Phenosane 23

RA 1010

Ralox 630

Sumilizer BP 101

Tetraalkofen BPE

Tetrakis [methylene (3,5-di-t-butyl-4-hydroxyhydrocinnamate)] methane

Tetrakis(3,5-di-tert-butyl-4-hydroxyhydrocinnamoyloxymethyl)methane

Tetrakis[methylene(3,5-di-tert-butyl-4-hydroxyhydrocinnamate)methane]

irganox 1010

pentaerythritol tetra(3,5-di-tert-butyl-4-hydroxyhydrocinnamate)

pentaerythritol tetrakis(3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionate)

tetrakis[methylene-3-(3,5-di-tert-butyl-4-hydroxyphenylpropionate)]methane

**Inchi:** InChI=1S/C73H108O12/c1-65(2,3)49-33-45(34-50(61(49)78)66(4,5)6)25-29-57(74)82-41

**InchiKey:** BGYHLZZASRKEJE-UHFFFAOYSA-N

**Formula:** C73H108O12

**SMILES:** CC(C)(C)c1cc(CCC(=O)OCC(COC(=O)CCc2cc(C(C)(C)C)c(O)c(C(C)(C)C)c2)(COC(=O)C

**Mol. weight [g/mol]:** 1177.63

**CAS:** 6683-19-8

# Physical Properties

Property code	Value	Unit	Source
gf	-592.22	kJ/mol	Joback Method
hf	-2462.88	kJ/mol	Joback Method
hfus	65.95	kJ/mol	Measurement and correlation of solid-liquid equilibria of Irganox 1010 with n-hexane
hfus	65.95	kJ/mol	Solubility of antioxidant 1010 in pure alkanols
hvap	269.51	kJ/mol	Joback Method
log10ws	-17.31		Crippen Method
logp	15.879		Crippen Method
mcvol	997.630	ml/mol	McGowan Method
pc	271.32	kPa	Joback Method
tb	2614.77	K	Joback Method
tc	10442.39	K	Joback Method
tf	1875.61	K	Joback Method
vc	3.553	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	11392.06	J/molxK	2614.77	Joback Method
cpg	887201.04	J/molxK	9137.79	Joback Method
cpg	525485.38	J/molxK	7833.19	Joback Method
cpg	279487.68	J/molxK	6528.58	Joback Method
cpg	126766.92	J/molxK	5223.98	Joback Method
cpg	44882.05	J/molxK	3919.37	Joback Method
cpg	1387075.69	J/molxK	10442.39	Joback Method
dvisc	0.0000000	Paxs	2614.77	Joback Method
dvisc	0.0000000	Paxs	2491.58	Joback Method
dvisc	0.0000000	Paxs	2368.38	Joback Method
dvisc	0.0000000	Paxs	2245.19	Joback Method
dvisc	0.0000000	Paxs	2122.00	Joback Method
dvisc	0.0000000	Paxs	1998.80	Joback Method
dvisc	0.0000000	Paxs	1875.61	Joback Method

# Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6683198&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6683198&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Solubility of antioxidant 1010 in pure alkanols:</b>	<a href="https://www.doi.org/10.1016/j.fluid.2008.11.005">https://www.doi.org/10.1016/j.fluid.2008.11.005</a>
<b>Measurement and correlation of solid-liquid equilibria of Irganox 1010</b>	<a href="https://www.doi.org/10.1016/j.fluid.2009.09.013">https://www.doi.org/10.1016/j.fluid.2009.09.013</a>
<b>Solubility of Irganox 1010 in (Alcohol + Water) Mixtures from (293.15 to 333.15) K:</b>	<a href="https://www.doi.org/10.1021/je900435t">https://www.doi.org/10.1021/je900435t</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dv<sub>isc</sub>:</b>	Dynamic viscosity
<b>g<sub>f</sub>:</b>	Standard Gibbs free energy of formation
<b>h<sub>f</sub>:</b>	Enthalpy of formation at standard conditions
<b>h<sub>fus</sub>:</b>	Enthalpy of fusion at standard conditions
<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10ws</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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