

# Benzeneethanamine, 2,5-dimethoxy-«alpha»,4-dimethyl-

<b>Other names:</b>	Phenethylamine, 2,5-dimethoxy-«alpha»,4-dimethyl-DOM STP 2,5-Dimethoxy-«alpha»,4-dimethylphenylethylamine 2,5-Dimethoxy-4-methylamphetamine 2,5-Dimethoxy-4-methylphenylisopropylamine 2,5-Dimethoxymethylamphetamine 2',5'-Dimethoxy-4'-methylamphetamine 4-Methyl-2,5-dimethoxyamphetamine (. +/-)-1-(2,5-Dimethoxy-4-methylphenyl)-2-aminopropane (. +/-)-1-(4-Methyl-2,5-dimethoxyphenyl)-2-aminopropane (. +/-)-2,5-Dimethoxy-4-methylamphetamine (. +/-)-DOM (RS)-DOM dl-2,5-Dimethoxy-4-methylamphetamine dl-4-Methyl-2,5-dimethoxyamphetamine STP (hallucinogen)
<b>Inchi:</b>	InChI=1S/C12H19NO2/c1-8-5-12(15-4)10(6-9(2)13)7-11(8)14-3/h5,7,9H,6,13H2,1-4H3
<b>InchiKey:</b>	NTJQREUGJKIARY-UHFFFAOYSA-N
<b>Formula:</b>	C12H19NO2
<b>SMILES:</b>	COc1cc(CC(C)N)c(OC)cc1C
<b>Mol. weight [g/mol]:</b>	209.28
<b>CAS:</b>	15588-95-1

## Physical Properties

Property code	Value	Unit	Source
gf	-12.31	kJ/mol	Joback Method
hf	-324.82	kJ/mol	Joback Method
hfus	23.76	kJ/mol	Joback Method
hvap	61.64	kJ/mol	Joback Method
log10ws	-2.95		Crippen Method
logp	1.902		Crippen Method
mcvol	177.900	ml/mol	McGowan Method
pc	2372.59	kPa	Joback Method
rinpol	1606.00		NIST Webbook
rinpol	1605.00		NIST Webbook
rinpol	1635.00		NIST Webbook

rinpol	1616.00		NIST Webbook
rinpol	1620.00		NIST Webbook
rinpol	1616.00		NIST Webbook
rinpol	1620.00		NIST Webbook
tb	632.51	K	Joback Method
tc	843.83	K	Joback Method
tf	401.70	K	Joback Method
vc	0.658	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	462.71	J/mol×K	632.51	Joback Method
cpg	478.13	J/mol×K	667.73	Joback Method
cpg	492.76	J/mol×K	702.95	Joback Method
cpg	506.60	J/mol×K	738.17	Joback Method
cpg	519.65	J/mol×K	773.39	Joback Method
cpg	531.90	J/mol×K	808.61	Joback Method
cpg	543.34	J/mol×K	843.83	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C15588951&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C15588951&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions

<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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