

pentyl hexanoate-d11

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|-----------------------------|---|
| Inchi: | InChI=1S/C11H22O2/c1-3-5-7-9-11(12)13-10-8-6-4-2/h3-10H2,1-2H3/i1D3,3D2,5D2,7D2 |
| InchiKey: | WRFZKAGPPQGDDQ-SBJQRNSFSA-N |
| Formula: | C11H11D11O2 |
| SMILES: | CCCCCOC(=O)CCCC |
| Mol. weight [g/mol]: | 197.36 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | -192.18 | kJ/mol | Joback Method |
| hf | -515.17 | kJ/mol | Joback Method |
| hfus | 27.03 | kJ/mol | Joback Method |
| hvap | 49.24 | kJ/mol | Joback Method |
| log10ws | -3.29 | | Crippen Method |
| logp | 3.300 | | Crippen Method |
| mcvol | 173.290 | ml/mol | McGowan Method |
| pc | 2014.51 | kPa | Joback Method |
| ripol | 1493.00 | | NIST Webbook |
| ripol | 1493.00 | | NIST Webbook |
| tb | 527.37 | K | Joback Method |
| tc | 698.36 | K | Joback Method |
| tf | 285.89 | K | Joback Method |
| vc | 0.675 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|-----------|---------|-----------------|---------------|
| cpg | 415.24 | J/molxK | 527.37 | Joback Method |
| cpg | 430.19 | J/molxK | 555.87 | Joback Method |
| cpg | 444.58 | J/molxK | 584.37 | Joback Method |
| cpg | 458.41 | J/molxK | 612.87 | Joback Method |
| cpg | 471.67 | J/molxK | 641.36 | Joback Method |
| cpg | 484.39 | J/molxK | 669.86 | Joback Method |
| cpg | 496.57 | J/molxK | 698.36 | Joback Method |
| dvisc | 0.0032030 | Paxs | 285.89 | Joback Method |

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|-------|-----------|------|--------|---------------|
| dvisc | 0.0015281 | Paxs | 326.14 | Joback Method |
| dvisc | 0.0008577 | Paxs | 366.38 | Joback Method |
| dvisc | 0.0005398 | Paxs | 406.63 | Joback Method |
| dvisc | 0.0003692 | Paxs | 446.88 | Joback Method |
| dvisc | 0.0002689 | Paxs | 487.12 | Joback Method |
| dvisc | 0.0002056 | Paxs | 527.37 | Joback Method |

Sources

| | |
|------------------------|---|
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=R329149&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci990307l |
| Crippen Method: | https://www.chemeo.com/doc/models/crippen_log10ws |

Legend

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|----------------------------|---|
| cp_g: | Ideal gas heat capacity |
| dvisc: | Dynamic viscosity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| hvap: | Enthalpy of vaporization at standard conditions |
| log₁₀ws: | Log ₁₀ of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m_{cvol}: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| ripol: | Polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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