

# [14]Helicene

<b>Other names:</b>	Diphenanthro[4,3-g:4',3'-g']naphtho[2,1-c:7,8-c']diphenanthrene
<b>Inchi:</b>	InChI=1S/C58H32/c1-3-7-46-33(5-1)9-11-35-13-15-37-17-19-39-21-23-41-25-27-43-29-3
<b>InchiKey:</b>	FYHJTLVERBIHKT-UHFFFAOYSA-N
<b>Formula:</b>	C58H32
<b>SMILES:</b>	c1ccc2c(c1)ccc1ccc3ccc4ccc5ccc6ccc7ccc8ccc9ccc%10ccc%11ccc%12ccc%13cccc%
<b>Mol. weight [g/mol]:</b>	728.87
<b>CAS:</b>	57483-71-3

## Physical Properties

Property code	Value	Unit	Source
gf	1820.78	kJ/mol	Joback Method
hf	1342.35	kJ/mol	Joback Method
hfus	96.60	kJ/mol	Joback Method
hvap	176.24	kJ/mol	Joback Method
ie	6.88	eV	NIST Webbook
log10ws	-24.93		Crippen Method
logp	16.678		Crippen Method
mcvol	551.340	ml/mol	McGowan Method
pc	840.16	kPa	Joback Method
tb	1859.62	K	Joback Method
tc	2342.70	K	Joback Method
tf	1345.18	K	Joback Method
vc	2.162	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	3456.59	J/molxK	1859.62	Joback Method
cpg	3840.73	J/molxK	1940.13	Joback Method
cpg	4281.43	J/molxK	2020.65	Joback Method
cpg	4783.25	J/molxK	2101.16	Joback Method
cpg	5350.77	J/molxK	2181.67	Joback Method
cpg	5988.57	J/molxK	2262.19	Joback Method
cpg	6701.21	J/molxK	2342.70	Joback Method

dvisc	0.0925601	Paxs	1345.18	Joback Method
dvisc	0.0872679	Paxs	1430.92	Joback Method
dvisc	0.0828278	Paxs	1516.66	Joback Method
dvisc	0.0790542	Paxs	1602.40	Joback Method
dvisc	0.0758107	Paxs	1688.14	Joback Method
dvisc	0.0729954	Paxs	1773.88	Joback Method
dvisc	0.0705303	Paxs	1859.62	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C57483713&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C57483713&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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