

# (E)-2-hexenal diethyl acetal

<b>Other names:</b>	(E)-1,1-diethoxyhex-2-ene
<b>Inchi:</b>	InChI=1S/C10H20O2/c1-4-7-8-9-10(11-5-2)12-6-3/h8-10H,4-7H2,1-3H3/b9-8+
<b>InchiKey:</b>	WMQKYHTZGYIHHD-CMDGGOBGS-A-N
<b>Formula:</b>	C10H20O2
<b>SMILES:</b>	CCCC=CC(OCC)OCC
<b>Mol. weight [g/mol]:</b>	172.26
<b>CAS:</b>	67746-30-9

## Physical Properties

Property code	Value	Unit	Source
gf	-98.90	kJ/mol	Joback Method
hf	-402.23	kJ/mol	Joback Method
hfus	20.71	kJ/mol	Joback Method
hvap	42.24	kJ/mol	Joback Method
log10ws	-2.64		Crippen Method
logp	2.742		Crippen Method
mcvol	159.200	ml/mol	McGowan Method
pc	2159.31	kPa	Joback Method
ripol	1285.00		NIST Webbook
ripol	1287.00		NIST Webbook
ripol	1285.00		NIST Webbook
ripol	1287.00		NIST Webbook
tb	476.76	K	Joback Method
tc	650.07	K	Joback Method
tf	226.84	K	Joback Method
vc	0.606	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	353.11	J/mol×K	476.76	Joback Method
cpg	367.77	J/mol×K	505.64	Joback Method
cpg	381.91	J/mol×K	534.53	Joback Method
cpg	395.52	J/mol×K	563.41	Joback Method

cpg	408.61	J/mol×K	592.30	Joback Method
cpg	421.20	J/mol×K	621.18	Joback Method
cpg	433.28	J/mol×K	650.07	Joback Method
dvisc	0.0044403	Paxs	226.84	Joback Method
dvisc	0.0015806	Paxs	268.49	Joback Method
dvisc	0.0007426	Paxs	310.15	Joback Method
dvisc	0.0004172	Paxs	351.80	Joback Method
dvisc	0.0002648	Paxs	393.45	Joback Method
dvisc	0.0001834	Paxs	435.11	Joback Method
dvisc	0.0001354	Paxs	476.76	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C67746309&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C67746309&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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