

5-Methyl-4-mercaptohexan-2-one

Inchi:	InChI=1S/C7H14OS/c1-5(2)7(9)4-6(3)8/h5,7,9H,4H2,1-3H3
InchiKey:	UQBDDFZDCXIBHM-UHFFFAOYSA-N
Formula:	C7H14OS
SMILES:	CC(=O)CC(S)C(C)C
Mol. weight [g/mol]:	146.25

Physical Properties

Property code	Value	Unit	Source
gf	-96.35	kJ/mol	Joback Method
hf	-272.47	kJ/mol	Joback Method
hfus	12.48	kJ/mol	Joback Method
hvap	43.88	kJ/mol	Joback Method
log10ws	-1.98		Crippen Method
logp	1.920		Crippen Method
mcvol	127.410	ml/mol	McGowan Method
pc	3276.53	kPa	Joback Method
rinpol	1107.00		NIST Webbook
rinpol	1069.00		NIST Webbook
rinpol	1069.00		NIST Webbook
rinpol	1069.00		NIST Webbook
ripol	1585.00		NIST Webbook
ripol	1585.00		NIST Webbook
tb	475.41	K	Joback Method
tc	682.94	K	Joback Method
tf	225.04	K	Joback Method
vc	0.475	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	262.97	J/mol×K	475.41	Joback Method
cpg	275.68	J/mol×K	510.00	Joback Method
cpg	287.76	J/mol×K	544.59	Joback Method
cpg	299.23	J/mol×K	579.18	Joback Method

cpg	310.10	J/mol×K	613.77	Joback Method
cpg	320.39	J/mol×K	648.35	Joback Method
cpg	330.12	J/mol×K	682.94	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R282496&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
ripol:	Polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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