

# (4aS,8R)-4a,8-Dimethyl-4,4a,5,6,7,8-hexahydronap

<b>Other names:</b>	6S,10R-Dimethylbicyclo[4.4.0]decan-1-en-3-one Bicyclo[4.4.0]decan-1-en-3-one, 6S,10R-dimethyl- 11,12,13-tris-nor-trans-Eudesm-5-en-7-one
<b>Inchi:</b>	InChI=1S/C12H18O/c1-9-4-3-6-12(2)7-5-10(13)8-11(9)12/h8-9H,3-7H2,1-2H3
<b>InchiKey:</b>	KLTKQWRPJDRMTL-UHFFFAOYSA-N
<b>Formula:</b>	C12H18O
<b>SMILES:</b>	CC1CCCC2(C)CCC(=O)C=C12
<b>Mol. weight [g/mol]:</b>	178.27
<b>CAS:</b>	69460-62-4

## Physical Properties

Property code	Value	Unit	Source
gf	15.51	kJ/mol	Joback Method
hf	-246.20	kJ/mol	Joback Method
hfus	8.75	kJ/mol	Joback Method
hvap	46.87	kJ/mol	Joback Method
log10ws	-3.29		Crippen Method
logp	3.102		Crippen Method
mcvol	155.490	ml/mol	McGowan Method
pc	2744.03	kPa	Joback Method
rinpol	1528.00		NIST Webbook
rinpol	1530.00		NIST Webbook
tb	576.72	K	Joback Method
tc	820.09	K	Joback Method
tf	352.20	K	Joback Method
vc	0.581	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	404.28	J/mol×K	576.72	Joback Method
cpg	425.04	J/mol×K	617.28	Joback Method
cpg	444.48	J/mol×K	657.84	Joback Method
cpg	462.77	J/mol×K	698.41	Joback Method

cpg	480.04	J/mol×K	738.97	Joback Method
cpg	496.46	J/mol×K	779.53	Joback Method
cpg	512.16	J/mol×K	820.09	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C69460624&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C69460624&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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