

Silhiperfola-4,7(14)-diene

Inchi:	InChI=1S/C15H22/c1-10-5-8-15-12(3)11(2)9-14(15,4)7-6-13(10)15/h9-10,13H,3,5-8H2,1
InchiKey:	WBXUOIDEYMLLNE-UHFFFAOYSA-N
Formula:	C15H22
SMILES:	C=C1C(C)=CC2(C)CCC3C(C)CCC132
Mol. weight [g/mol]:	202.34
CAS:	210637-49-3

Physical Properties

Property code	Value	Unit	Source
gf	288.19	kJ/mol	Joback Method
hf	-6.16	kJ/mol	Joback Method
hfus	12.96	kJ/mol	Joback Method
hvap	47.57	kJ/mol	Joback Method
log10ws	-4.53		Crippen Method
logp	4.335		Crippen Method
mcvol	181.030	ml/mol	McGowan Method
pc	2222.89	kPa	Joback Method
rinpol	1361.00		NIST Webbook
rinpol	1362.50		NIST Webbook
tb	570.47	K	Joback Method
tc	798.02	K	Joback Method
tf	376.11	K	Joback Method
vc	0.696	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	482.27	J/mol×K	570.47	Joback Method
cpg	503.46	J/mol×K	608.40	Joback Method
cpg	523.18	J/mol×K	646.32	Joback Method
cpg	541.72	J/mol×K	684.25	Joback Method
cpg	559.40	J/mol×K	722.17	Joback Method
cpg	576.51	J/mol×K	760.10	Joback Method
cpg	593.35	J/mol×K	798.02	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C210637493&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpola:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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