

# 1,1,6-trimethyl-3-methylene-2-(3,6,10,13,14-pentamethyl-3-vinyl-4-pentadecyl)ethane

<b>Other names:</b>	(2R,6S)-1,1,6-Trimethyl-3-methylene-2-[(E)-3,6,10,13,14-pentamethyl-3-vinyl-4-pentadecyl]ethane
<b>Inchi:</b>	InChI=1S/C32H58/c1-12-32(11,23-21-30-28(7)18-19-29(8)31(30,9)10)22-20-26(5)15-13-14
<b>InchiKey:</b>	MPEPBAMJSYOXJV-LSDHQDQOSA-N
<b>Formula:</b>	C32H58
<b>SMILES:</b>	<chem>C=CC(C)(C=CC(C)CCCC(C)CCC(C)C(C)C)CCC1C(=C)CCC(C)C1(C)C</chem>
<b>Mol. weight [g/mol]:</b>	442.80

## Physical Properties

Property code	Value	Unit	Source
gf	436.32	kJ/mol	Joback Method
hf	-377.91	kJ/mol	Joback Method
hfus	42.57	kJ/mol	Joback Method
hvap	82.08	kJ/mol	Joback Method
log10ws	-10.74		Crippen Method
logp	10.658		Crippen Method
mvol	437.980	ml/mol	McGowan Method
pc	643.53	kPa	Joback Method
rinpol	2669.00		NIST Webbook
tb	937.02	K	Joback Method
tc	1148.18	K	Joback Method
tf	422.46	K	Joback Method
vc	1.667	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1538.40	J/molxK	937.02	Joback Method
cpg	1566.74	J/molxK	972.21	Joback Method
cpg	1594.33	J/molxK	1007.41	Joback Method
cpg	1621.37	J/molxK	1042.60	Joback Method
cpg	1648.05	J/molxK	1077.80	Joback Method
cpg	1674.55	J/molxK	1112.99	Joback Method
cpg	1701.07	J/molxK	1148.18	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U373940&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U373940&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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