

# trans-p-(1-Butenyl)-anisole

<b>Inchi:</b>	InChI=1S/C11H14O/c1-3-4-5-10-6-8-11(12-2)9-7-10/h4-9H,3H2,1-2H3/b5-4+
<b>InchiKey:</b>	XDQBFRZHMRLPGN-SNAWJCMRSA-N
<b>Formula:</b>	C11H14O
<b>SMILES:</b>	CCC=Cc1ccc(OC)cc1
<b>Mol. weight [g/mol]:</b>	162.23
<b>CAS:</b>	61866-52-2

## Physical Properties

Property code	Value	Unit	Source
gf	119.74	kJ/mol	Joback Method
hf	-60.31	kJ/mol	Joback Method
hfus	19.29	kJ/mol	Joback Method
hvap	45.39	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	3.118		Crippen Method
mcvol	143.660	ml/mol	McGowan Method
pc	2704.22	kPa	Joback Method
rinpol	1270.00		NIST Webbook
rinpol	1270.00		NIST Webbook
tb	509.32	K	Joback Method
tc	720.42	K	Joback Method
tf	269.82	K	Joback Method
vc	0.541	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	308.52	J/molxK	509.32	Joback Method
cpg	374.99	J/molxK	685.23	Joback Method
cpg	363.20	J/molxK	650.05	Joback Method
cpg	350.68	J/molxK	614.87	Joback Method
cpg	337.42	J/molxK	579.69	Joback Method
cpg	323.37	J/molxK	544.50	Joback Method
cpg	386.09	J/molxK	720.42	Joback Method

dvisc	0.0001503	Paxs	509.32	Joback Method
dvisc	0.0001907	Paxs	469.40	Joback Method
dvisc	0.0002530	Paxs	429.49	Joback Method
dvisc	0.0003556	Paxs	389.57	Joback Method
dvisc	0.0005401	Paxs	349.65	Joback Method
dvisc	0.0009139	Paxs	309.74	Joback Method
dvisc	0.0018064	Paxs	269.82	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C61866522&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C61866522&amp;Units=SI</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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