

# 1-Dotriacontene

<b>Other names:</b>	dotriacontan-1-ene
<b>Inchi:</b>	InChI=1S/C32H64/c1-3-5-7-9-11-13-15-17-19-21-23-25-27-29-31-32-30-28-26-24-22-20
<b>InchiKey:</b>	WMRDPJYQERQCEP-UHFFFAOYSA-N
<b>Formula:</b>	C32H64
<b>SMILES:</b>	C=CCCCCCCCCCCCCCCCCCCCCCCCCCCCCCCC
<b>Mol. weight [g/mol]:</b>	448.85
<b>CAS:</b>	18435-55-7

## Physical Properties

Property code	Value	Unit	Source
gf	306.40	kJ/mol	Joback Method
hf	-578.38	kJ/mol	Joback Method
hfus	77.36	kJ/mol	Joback Method
hvap	86.16	kJ/mol	Joback Method
log10ws	-13.07		Crippen Method
logp	12.505		Crippen Method
mcvol	457.440	ml/mol	McGowan Method
pc	554.15	kPa	Joback Method
rinpol	3188.00		NIST Webbook
rinpol	3188.00		NIST Webbook
tb	928.24	K	Joback Method
tc	1151.65	K	Joback Method
tf	448.64	K	Joback Method
vc	1.808	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1736.68	J/molxK	1151.65	Joback Method
cpg	1715.39	J/molxK	1114.41	Joback Method
cpg	1692.87	J/molxK	1077.18	Joback Method
cpg	1668.99	J/molxK	1039.94	Joback Method
cpg	1643.63	J/molxK	1002.71	Joback Method
cpg	1616.68	J/molxK	965.47	Joback Method

cpg	1588.01	J/mol×K	928.24	Joback Method
dvisc	0.0008223	Paxs	448.64	Joback Method
dvisc	0.0000193	Paxs	928.24	Joback Method
dvisc	0.0000269	Paxs	848.31	Joback Method
dvisc	0.0000400	Paxs	768.37	Joback Method
dvisc	0.0000655	Paxs	688.44	Joback Method
dvisc	0.0001220	Paxs	608.51	Joback Method
dvisc	0.0002742	Paxs	528.57	Joback Method
hvapt	119.80	kJ/mol	606.00	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.51987e+01
Coeff. B	-6.28750e+03
Coeff. C	-1.46356e+02
Temperature range (K), min.	568.02
Temperature range (K), max.	782.27

## Sources

The Yaws Handbook of Vapor Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>  
<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C18435557&Units=SI>

## Legend

**cpg:** Ideal gas heat capacity  
**dvisc:** Dynamic viscosity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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