

# Acenaphthylene, dodecahydro-5-pentadecyl-

**Other names:**

5-n-Pentadecyl-(decahydroacenaphthylene)  
5-n-Pentadecyl-(dodecahydroacenaphthene)  
5-n-Pentadecyltricyclo(6.3.1.0(4,12))dodecane  
5-Pentadecyldodecahydroacenaphthalene

**Inchi:** InChI=1S/C27H50/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-16-23-19-20-25-22-21-24-17-15-1**InchiKey:** JSWHXZRXLGIVGRL-UHFFFAOYSA-N**Formula:** C27H50**SMILES:** CCCCCCCCCCCCCCCC1CCC2CCC3CCCC1C32**Mol. weight [g/mol]:** 374.69**CAS:** 55282-69-4

## Physical Properties

Property code	Value	Unit	Source
gf	306.99	kJ/mol	Joback Method
hf	-441.37	kJ/mol	Joback Method
hfus	55.93	kJ/mol	Joback Method
hvap	75.33	kJ/mol	Joback Method
log10ws	-9.60		Crippen Method
logp	9.320		Crippen Method
mcvol	358.710	ml/mol	McGowan Method
pc	859.48	kPa	Joback Method
tb	840.85	K	Joback Method
tc	1036.16	K	Joback Method
tf	428.83	K	Joback Method
vc	1.393	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1258.64	J/mol×K	840.85	Joback Method
cpg	1283.93	J/mol×K	873.40	Joback Method
cpg	1307.85	J/mol×K	905.95	Joback Method
cpg	1330.50	J/mol×K	938.50	Joback Method
cpg	1351.97	J/mol×K	971.06	Joback Method

cpg	1372.36	J/mol×K	1003.61	Joback Method
cpg	1391.77	J/mol×K	1036.16	Joback Method
dvisc	0.0032449	Paxs	428.83	Joback Method
dvisc	0.0020348	Paxs	497.50	Joback Method
dvisc	0.0014289	Paxs	566.17	Joback Method
dvisc	0.0010831	Paxs	634.84	Joback Method
dvisc	0.0008667	Paxs	703.51	Joback Method
dvisc	0.0007215	Paxs	772.18	Joback Method
dvisc	0.0006190	Paxs	840.85	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282694&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55282694&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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