

# Benzene, 1-(1-heptyldodecyl)-4-methyl-

<b>Other names:</b>	1-Methyl-4-(1'-n-heptyldodecyl)benzene 8-(4'-Methylphenyl)nonadecane 8-(4-Tolyl)nonadecane 8-p-Tolylnonadecane
<b>Inchi:</b>	InChI=1S/C26H46/c1-4-6-8-10-11-12-13-15-17-19-25(18-16-14-9-7-5-2)26-22-20-24(3)2
<b>InchiKey:</b>	OGAMHWGABJDLOK-UHFFFAOYSA-N
<b>Formula:</b>	C26H46
<b>SMILES:</b>	<chem>CCCCCCCCCCC(CCCCCC)c1ccc(C)cc1</chem>
<b>Mol. weight [g/mol]:</b>	358.64
<b>CAS:</b>	55191-36-1

## Physical Properties

Property code	Value	Unit	Source
gf	268.38	kJ/mol	Joback Method
hf	-360.19	kJ/mol	Joback Method
hfus	53.23	kJ/mol	Joback Method
hvap	76.02	kJ/mol	Joback Method
log10ws	-9.83		Crippen Method
logp	9.360		Crippen Method
mcvol	353.440	ml/mol	McGowan Method
pc	870.16	kPa	Joback Method
tb	825.50	K	Joback Method
tc	1015.12	K	Joback Method
tf	281.85 ± 0.50	K	NIST Webbook
vc	1.377	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1247.43	J/molxK	1015.12	Joback Method
cpg	1133.18	J/molxK	825.50	Joback Method
cpg	1154.99	J/molxK	857.10	Joback Method
cpg	1175.61	J/molxK	888.71	Joback Method
cpg	1195.11	J/molxK	920.31	Joback Method

cpg	1213.54	J/mol×K	951.91	Joback Method
cpg	1230.96	J/mol×K	983.51	Joback Method
dvisc	0.0000414	Paxs	825.50	Joback Method
dvisc	0.0013255	Paxs	406.72	Joback Method
dvisc	0.0004873	Paxs	476.52	Joback Method
dvisc	0.0002313	Paxs	546.31	Joback Method
dvisc	0.0001300	Paxs	616.11	Joback Method
dvisc	0.0000822	Paxs	685.91	Joback Method
dvisc	0.0000565	Paxs	755.70	Joback Method
hvapt	94.50	kJ/mol	499.50	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.34346e+01
Coeff. B	-5.03108e+03
Coeff. C	-1.28522e+02
Temperature range (K), min.	511.20
Temperature range (K), max.	747.87

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C55191361&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C55191361&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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