

# Clomiphene

**Other names:**

Clomifene  
Ethanamine, 2-[4-(2-chloro-1,2-diphenylethenyl)phenoxy]-N,N-diethyl-  
Triethylamine, 2-(p-(2-chloro-1,2-diphenylvinyl)phenoxy)-  
2-(4-(2-Chloro-1,2-diphenylethenyl)phenoxy)-N,N-diethylethanamine  
2-(p-(«beta»-Chloro-«alpha»-phenylstyryl)phenoxy)-triethylamine  
Cisclomiphene  
Clomiphene B

**Inchi:** InChI=1S/C26H28ClNO/c1-3-28(4-2)19-20-29-24-17-15-22(16-18-24)25(21-11-7-5-8-12-**InchiKey:** GKIRPKYJQBWNGO-QPLCGJKRSA-N**Formula:** C26H28ClNO**SMILES:** CCN(CC)CCOc1ccc(C(=C(Cl)c2ccccc2)c2ccccc2)cc1**Mol. weight [g/mol]:** 405.96**CAS:** 911-45-5

## Physical Properties

Property code	Value	Unit	Source
gf	552.61	kJ/mol	Joback Method
hf	135.36	kJ/mol	Joback Method
hfus	50.82	kJ/mol	Joback Method
hvap	89.92	kJ/mol	Joback Method
log10ws	-7.22		Crippen Method
logp	6.563		Crippen Method
mcvol	329.710	ml/mol	McGowan Method
pc	1341.76	kPa	Joback Method
rinpol	2930.00		NIST Webbook
tb	955.51	K	Joback Method
tc	1198.20	K	Joback Method
tf	526.18	K	Joback Method
vc	1.234	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1017.81	J/mol×K	955.51	Joback Method
cpg	1034.18	J/mol×K	995.96	Joback Method
cpg	1049.40	J/mol×K	1036.41	Joback Method
cpg	1063.61	J/mol×K	1076.85	Joback Method
cpg	1076.96	J/mol×K	1117.30	Joback Method
cpg	1089.63	J/mol×K	1157.75	Joback Method
cpg	1101.75	J/mol×K	1198.20	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C911455&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C911455&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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