

# Fumaric acid, hexadecyl 2,4,6-trichlorophenyl ester

Inchi:	InChI=1S/C26H37Cl3O4/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-18-32-24(30)16-17-25(3
InchiKey:	OERIBWCXCWLROJ-WUKNDPDISA-N
Formula:	C26H37Cl3O4
SMILES:	CCCCCCCCCCCCCCCCOC(=O)C=CC(=O)Oc1c(Cl)cc(Cl)cc1Cl
Mol. weight [g/mol]:	519.93

## Physical Properties

Property code	Value	Unit	Source
gf	-171.85	kJ/mol	Joback Method
hf	-797.45	kJ/mol	Joback Method
hfus	74.34	kJ/mol	Joback Method
hvap	109.16	kJ/mol	Joback Method
log10ws	-10.09		Crippen Method
logp	9.133		Crippen Method
mcvol	400.740	ml/mol	McGowan Method
pc	867.60	kPa	Joback Method
rinsol	3547.00		NIST Webbook
tb	1104.93	K	Joback Method
tc	1357.88	K	Joback Method
tf	675.76	K	Joback Method
vc	1.558	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1283.94	J/molxK	1104.93	Joback Method
cpg	1297.75	J/molxK	1147.09	Joback Method
cpg	1310.03	J/molxK	1189.25	Joback Method
cpg	1320.85	J/molxK	1231.41	Joback Method
cpg	1330.31	J/molxK	1273.57	Joback Method
cpg	1338.49	J/molxK	1315.72	Joback Method
cpg	1345.49	J/molxK	1357.88	Joback Method
dvisc	0.0001131	Paxs	675.76	Joback Method
dvisc	0.0000654	Paxs	747.29	Joback Method

dvisc	0.0000417	Paxs	818.82	Joback Method
dvisc	0.0000285	Paxs	890.34	Joback Method
dvisc	0.0000206	Paxs	961.87	Joback Method
dvisc	0.0000156	Paxs	1033.40	Joback Method
dvisc	0.0000123	Paxs	1104.93	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U348279&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U348279&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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